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=> d his full
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(FILE 'HOME' ENTERED AT 10:38:48 ON 02 MAY 2005)
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                  E AUDENAERT F/AU
                  SEA ABB=ON PLU=ON ("AUDENAERT F"/AU OR "AUDENAERT FRANS"/AU OR "AUDENAERT FRANS ALBERT"/AU)
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                   E DAMS R/AU
              239 SEA ABB=ON PLU=ON ("DAMS R"/AU OR "DAMS R A J"/AU OR "DAMS RALPH W L J"/AU)
L2
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               41 SEA ABB=ON PLU=ON ("DAMS RUDOLF"/AU OR "DAMS RUDOLF J"/AU OR "DAMS RUDOLF J E A"/AU OR "DAMS RUDOLF JOZEF"/AU OR "DAMS RUDOLPH J"/AU)
L3
                   E TAN L/AU
              140 SEA ABB=ON PLU=ON ("TAN L"/AU OR "TAN L S"/AU OR "TAN L S K"/AU OR "TAN L SENG"/AU)
L4
                   E TAN LIAN/AU
                7 SEA ABB=ON PLU=ON ("TAN LIAN"/AU OR "TAN LIAN S"/AU OR "TAN
L5
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             2229 SEA ABB=ON PLU=ON (3M (1A) INNOV?)/CS, PA
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                   E POLYETHERS/CT
                  E SURFACTANT/CT
                   E SURFACTANTS/CT
                  E E3+ALL
                   QUE ABB=ON PLU=ON SURFACTANTS+OLD, NT/CT
L7
                  E E35
                   E E3+ALL
             5960 SEA ABB=ON PLU=ON SURFACE ACTIVITY+OLD, NT/CT
L8
                   E SURFACTANT ADSORPTION/CT
                   E E3+ALL
              414 SEA ABB=ON PLU=ON SURFACTANT ADSORPTION/CT
L9
                   E PENETRATING AGENTS/CT
                   E E3+ALL
L10
              572 SEA ABB=ON PLU=ON PENETRATING AGENTS/CT
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L11
                   OR L9 OR L10)
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L13
               13 SEA ABB=ON PLU=ON L11 AND ?FLUOR?
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                  TRA L13 1- RN:
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L14
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              226 SEA ABB=ON PLU=ON L14
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L22
=> b hcap
FILE 'HCAPLUS' ENTERED AT 10:52:34 ON 02 MAY 2005
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FILE COVERS 1907 - 2 May 2005 VOL 142 ISS 19 FILE LAST UPDATED: 1 May 2005 (20050501/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

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ANSWER 1 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     2003:818177 HCAPLUS
AN
DN
     139:324175
ED
     Entered STN: 17 Oct 2003
ΤI
     Dispersions containing bicomponent fluoropolymer particles and
     use thereof
IN
     Buckanin, Richard S.; Tan, Lian S.; McAlister, E. Steven
     3M Innovative Properties Company, USA
PA
S0
     U.S. Pat. Appl. Publ., 13 pp.
     CODEN: USXXCO
DΤ
     Patent
     English
LA
     ICM C08F116-12
INCL 526247000; 526249000; 526250000; 526206000; 524458000; 524805000
     37-3 (Plastics Manufacture and Processing)
FAN. CNT 1
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                                  DATE
                                               APPLICATION NO.
                                                                        DATE
                                  20031016
                                               US 2002-117714
                                                                        20020405
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     US 2003195314
                            A1
     US 6822059
                            B2
                                  20041123
     WO 2003087179
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                                               WO 2003-US4070
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                                  20050105
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PRAI US 2002-117714
                                  20020405
                            Α
     WO 2003-US4070
                                  20030212
CLASS
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 US 2003195314
                  ICM
                          C08F116-12
                          526247000; 526249000; 526250000; 526206000; 524458000;
                  INCL
                          524805000
 US 2003195314
                  NCL
                          526/247.000; 524/458.000; 524/805.000; 526/206.000;
                          526/249.000; 526/250.000
                  ECLA
                          C08F002/24; C08F216/14B; C08F259/08; C08F261/06;
                          C08L027/12+B2; C08L051/00B+B; D06M015/353
                          C08F002/24; C08F216/14B; C08F259/08; C08F261/06;
 WO 2003087179
                  ECLA
                          C08L027/12+B2; C08L051/00B+B; D06M015/353
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A method of making a bicomponent fluoropolymer dispersion
     comprises: a. pre-emulsifying an aqueous mixture of a perfluorovinyl
     ether monomer in the presence of a fluorochem. emulsifier to an average emulsion droplet size of one micron or less, and b. polymerizing the
     perfluorovinyl ether in the presence of a free-radical initiator
     at temperature and for a time sufficient to produce particles of poly(
     perfluorovinyl ether), c. subsequently adding at least one addnl.
fluorinated comonomer without addnl. emulsifier, and d. further
polymerizing the resulting mixture  The dispersions of the present invention may
     be used for rendering fibrous substrates oil repellent, water repellent
     and/or stain repellent without altering the looks and feel of the
     substrate. A dispersion was prepared by polymerization of
     perfluoropropoxypropyl vinyl ether followed by polymerization with
      hexafluoropropene and vinylidene fluoride.
     fluoropolymer dispersion emulsion polymn oil water repellent
     Emulsification
     Oilproofing agents
      Water-resistant materials
         (dispersions containing bicomponent fluoropolymer particles and
         use thereof)
     Fluoropolymers, uses
ΙT
     RL: POF (Polymer in formulation); TEM (Technical or engineered material
     use); USES (Uses)
         (dispersions containing bicomponent fluoropolymer particles and
         use thereof)
     Polymerization
         (emulsion; dispersions containing bicomponent fluoropolymer
         particles and use thereof)
IT
     Emulsifying agents
         (fluorochem.; dispersions containing bicomponent
         fluoropolymer particles and use thereof)
IT
     Textiles
         (substrate; dispersions containing bicomponent fluoropolymer
         particles and use thereof)
IT 196623-59-3, FLUOROLINK C 314057-81-3, NAFION SE10172 RL: MOA (Modifier or additive use); USES (Uses)
         (dispersing agent; dispersions containing bicomponent fluoropolymer
         particles and use thereof)
     612801-40-8P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
         (dispersions containing bicomponent fluoropolymer particles and
         use thereof)
     ANSWER 2 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
     2003:796358 HCAPLUS
     139:292888
ED
     Entered STN: 10 Oct 2003
     Dispersions containing perfluorovinyl ether homopolymers useful
ΤI
     as textile finishing materials
     Tan, Lian S.; Buckanin, Richard S.; McAlister, E. Steven
IN
PA
      3M Innovative Properties Company, USA
S<sub>0</sub>
     U.S. Pat. Appl. Publ., 12 pp.
     CODEN: USXXCO
DΤ
     Patent
LA
     English
     ICM C08F116-12
INCL 526247000; 526249000; 526250000; 526206000; 524458000; 524805000
     37-3 (Plastics Manufacture and Processing)
     Section cross-reference(s): 40
FAN. CNT 1
     PATENT NO.
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                            KIND
                                    DATE
                                                  APPLICATION NO.
PΙ
     US 2003191258
                             A1
                                    20031009
                                                  US 2002-117715
                                                                           20020405
                                    20041221
      US 6833418
                             B2
      WO 2003087176
                                    20031023
                                                  WO 2003-US3257
                                                                           20030204
                             A1
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                                              JP, KE, KG, KP, KR, KZ, LC, LK, LR,
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MK, MN, MW, MX, MZ, NO, NZ, OM, PH,

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                                            EP 2003-706061
     EP 1492826
                                20050105
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     BR 2003009036
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                                            BR 2003-9036
                                                                    20030204
                          Α
PRAI US 2002-117715
                                20020405
     WO 2003-US3257
                                20030204
CLASS
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                 CLASS
                        PATENT FAMILY CLASSIFICATION CODES
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                 ICM
                        C08F116-12
                 INCL
                        526247000; 526249000; 526250000; 526206000; 524458000;
                        524805000
 US 2003191258
                 NCL
                        526/247.000; 524/458.000; 524/805.000; 526/206.000;
                        526/249,000; 526/250,000
                 ECLA
                        C08F116/14
AB
     A method for producing a perfluorovinyl ether homopolymer
     dispersion comprises the steps of (a) pre-emulsifying an aqueous mixture of a
     perfluorovinyl ether in the presence of a fluorinated
     emulsifier to an average emulsion droplet size of one micron or less, and (b)
     polymerizing the perfluorovinyl ether in the presence of an initiator
     at temperature and for time sufficient to produce particles of poly(
     perfluorovinyl ether). The aqueous fluoropolymer dispersion
     can comprise bicomponent particles of the poly(perfluorovinyl
     ether) and a second fluoropolymer. In combination with
     auxiliary textile finishing agents, the poly(perfluorovinyl
     ether) dispersions can be used for improving oil repellency, soil/stain
     repellency and/or water repellency of fibrous substrates.
     Fluorolink C (3.3) and potassium hydroxide (0.186) were dissolved
     in deionized water (90.0), followed by addition of Nafion SE 10172 (0.0175)
     and perfluoro(propoxypropyl vinyl ether) (50.0 g). The
     resulting aqueous mixture was sonicated for 60 s and homogenized to yield an
     emulsion with a mean droplet size of 144 nm. A solution of deionized water
     (10.0), sodium bicarbonate (0.2) and ammonium persulfate (0.2 g) was
     added, and the emulsion was heated at 60° for 20 h under nitrogen
     producing a homopolymer dispersion with a mean particle size of 62 nm, the
     monomer conversion being 82%.
     perfluorovinyl ether polymer dispersion textile finishing agent
     Fabric finishing
        (agents; dispersions containing perfluorovinyl ether homopolymers
        useful as textile finishing materials)
ΙT
     Disperse systems
     Oilproofing agents
     Soilproofing agents
     Textiles
     Waterproofing agents
        (dispersions containing perfluorovinyl ether homopolymers useful
        as textile finishing materials)
     Fluoropolymers, preparation
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (dispersions containing perfluorovinyl ether homopolymers useful
        as textile finishing materials)
IT
     Ionomers
     RL: NUU (Other use, unclassified); USES (Uses)
        (emulsifiers; production of dispersions containing perfluorovinyl
        ether homopolymers by emulsion polymerization)
IT
     Polymerization
         (emulsion; production of dispersions containing perfluorovinyl ether
        homopolymers by)
ΙT
     Polyethers, uses
     RL: NUU (Other use, unclassified); USES (Uses)
        (perfluoro, emulsifiers; production of dispersions containing
        perfluorovinyl ether homopolymers by emulsion polymerization)
     Fluoropolymers, uses
IT
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RL: NUU (Other use, unclassified); USES (Uses)
        (polyether-, perfluoro, emulsifiers; production of dispersions
        containing perfluorovinyl ether homopolymers by emulsion polymerization)
IT
     Emulsifying agents
        (production of dispersions containing perfluorovinyl ether
        homopolymers by emulsion polymerization)
     70087-25-1P, Perfluoropropyl vinyl ether homopolymer 98973-10-5P 384819-14-1P
                   384819-14-1P
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (dispersions containing perfluorovinyl ether homopolymers useful
        as textile finishing materials)
     29420-49-3, Potassium perfluorobutanesulfonate
                                                           196623-59-3,
     Fluorolink C 314057-81-3, Nafion SE 10172
RL: NUU (Other use, unclassified); USES (Uses)
        (emulsifier; production of dispersions containing perfluorovinyl
        ether homopolymers by emulsion polymerization)
     7727-54-0, Ammonium persulfate RL: CAT (Catalyst use); USES (Uses)
ΙT
        (production of dispersions containing perfluorovinyl ether
        homopolymers by emulsion polymerization)
    ANSWER 3 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     2003:757353 HCAPLUS
AN
DN
     139:261660
ED
     Entered STN: 26 Sep 2003
     Emulsion polymerization of using fluorinated surfactants
     Tan, Lian S.; Buckanin, Richard S.
IN
PA
         Pat. Appl. Publ., 24 pp., Cont.-in-part of U.S. Ser. No. 861, 782.
S<sub>0</sub>
     U.S.
     CODEN: USXXCO
DT
     Patent
     English
LA
     ICM C08F114-18
INCL 524502000; 524529000; 524543000; 526242000; 526245000
     35-4 (Chemistry of Synthetic High Polymers)
FAN. CNT 2
     PATENT NO.
                          KIND
                                  DATE
                                               APPLICATION NO.
                                                                       DATE
                                                                        20030124
                                  20030925
                                               US 2003-350457
PΙ
     US 2003181572
                            A1
     US 2003139521
                            A1
                                  20030724
                                               US 2001-861782
                                                                        20010521
     US 6737489
                                  20040518
                            B2
     WO 2004067588
                                  20040812
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                                  20030124
     US 2003-350457
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                  ICM
                          C08F114-18
                          524502000; 524529000; 524543000; 526242000; 526245000
                  INCL
                          524/502.000; 524/529.000; 524/543.000; 526/242.000;
 US 2003181572
                  NCL
                          526/245.000
                  ECLA
                         C08L027/12+B2; D06M013/395; D06M013/432; D06M015/256;
                          D06M015/263; D06M015/277; D06M015/353; D06M015/356N;
                          D06M015/356S; D06M015/568
 US 2003139521
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                          526/242.000; 524/502.000; 524/529.000; 524/543.000;
                          526/247.000
                         CO8L027/12+B2; D06M013/395; D06M013/432; D06M015/256;
                  ECLA
                          D06M015/263; D06M015/277; D06M015/353; D06M015/356N;
                          D06M015/356S; D06M015/568
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WO 2004067588
                ECLA C08F014/18+2/22
     The method comprises adding a fluorinated surfactant having a
     mol. weight of at least about 1000 g/mol to an emulsion polymerization process.
     Ammonium perfluorooctanoate was used as an emulsifier in polymerization
     of CF2:CF2 and perfluoro (Me vinyl) ether.
     fluorinated surfactant emulsion polymn
ST
ΙT
     Polymerization
         (emulsion; emulsion polymerization of using fluorinated surfactants)
ΙT
     Surfactants
         (fluorinated; emulsion polymerization of using fluorinated
        surfactants)
IT
     Polyethers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
         (perfluoro, surfactant; emulsion polymerization of using
        fluorinated surfactants)
     Fluoropolymers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polyether-, perfluoro, surfactant; emulsion polymerization of using fluorinated surfactants)
     26425-79-6P
                   70087-25-1P
                                  74499-68-6P
                                                 80975-16-2P
                                                                98973-10-5P
     349118-39-4P
                    384819-14-1P
                                    477198-48-4P .
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (emulsion polymerization of using fluorinated surfactants)
     3825-26-1, Ammonium perfluorooctanoate
     RL: TEM (Technical or engineered material use); USES (Uses)
         (emulsion polymerization of using fluorinated surfactants)
     314057-81-3, Nafion SE10172
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
         (surfactant; emulsion polymerization of using fluorinated
        surfactants)
     ANSWER 4 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     2003:417794 HCAPLUS
AN
     139:7655
DN
ED
     Entered STN: 01 Jun 2003
     Compositions for aqueous delivery of fluorinated silanes
·TI
IN
     Terrazas, Michael S.; Pellerite, Mark J.; Dams, Rudolf J.
PA
     3M Innovative Properties Company, USA
     PCT Int. Appl., 43 pp.
S<sub>0</sub>
     CODEN: PIXXD2
DT
     Patent
     English
LA
IC
     ICM C08G065-336
     ICS C08G077-00; C09D183-00
     37-3 (Plastics Manufacture and Processing)
     Section cross-reference(s): 42
FAN. CNT 1
     PATENT NO.
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                                               APPLICATION NO.
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     WO 2003044075
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     US 6592659
                           B1
                                  20030715
                                              US 2001-2543
                                                                       20011115
                                  20040811
                                              EP 2002-776307
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     EP 1444290
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     IP 2005509708
                           T2
                                  20050414
                                               JP 2003-545709
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PRAI US 2001-2543
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CLASS
 PATENT NO.
                  CLASS
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WO 2003044075
                   ICM
                            C08G065-336
                            C08G077-00; C09D183-00
                    ICS
                   NCL
                            106/287. 130; 106/287. 140; 106/287. 270; 427/387. 000;
 US 6592659
                            428/421,000; 428/447,000
                           4H020/BA11; 4H020/BA21; 4H020/BA23; 4H020/BA36;
 JP 2005509708
                   FTERM
                           4J002/BQ002; 4J002/CH052; 4J002/CP081; 4J002/CP181; 4J002/EF006; 4J002/EN006; 4J002/EV286; 4J002/FD312;
                            4J002/FD316; 4J002/GH01; 4J005/AA04; 4J005/BD05;
                           4J005/BD08; 4J038/CG032; 4J038/CG142; 4J038/CH262; 4J038/CJ252; 4J038/DF011; 4J038/DF012; 4J038/DL051; 4J038/DL061; 4J038/DL161; 4J038/GA02; 4J038/GA06;
                           4J038/GA09; 4J038/GA12; 4J038/GA13; 4J038/HA156; 4J038/KA09; 4J038/MA07; 4J038/MA08; 4J038/MA09;
                            4J038/NA07
AB
     The invention relates to a dilutable, nonaq. concentrate and an aqueous dilution used
     for aqueous delivery of fluorinated silanes to a substrate, a method
     of treating a substrate with the aqueous dilution composition to render it oil and
     water repellent, and articles having coatings made from the aqueous dilution The
     aqueous dilution may be coated on a substrate to provide a durable coating.
     Thus, shaking 5.0 g a 10% solution of Krytox 157FS(L) (carboxy group-containing
     fluoropolymer) ammonium salt in MeOH with 1.0 g
      trimethoxysilylpropylamido group-terminated perfluorinated
     polyether (I) gave a single-phase liquid containing 16.7% I.
                                                                            Diluting 0.5 g the
     concentrated solution above with 10.0 g water gave a clear dilution with no precipitation
     silane deriv perfluorinated polyether aq conc diln surface
      treatment
IT
     Ceramics
     Coating materials
         (dilutable concs. for aqueous delivery of fluorinated silanes)
     Glass, miscellaneous
     RL: MSC (Miscellaneous)
         (dilutable concs. for aqueous delivery of fluorinated silanes)
IT
     Surfactants
         (fluorosurfactants; dilutable concs. for aqueous delivery of
         fluorinated silanes)
IT
     Polyethers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
         (perfluoro, silylated; dilutable concs. for aqueous delivery of
         fluorinated silanes)
IT
     Fluoropolymers, uses
      RL: TEM (Technical or engineered material use); USES (Uses)
         (polyether-, perfluoro, silylated; dilutable concs. for aqueous delivery of fluorinated silanes)
      101947-16-4
                     220864-25-5
                                     233676-28-3
     RL: TEM (Technical or engineered material use); USES (Uses)
         (surface treatment; dilutable concs. for aqueous delivery of
         fluorinated silanes)
                                                  126600-08-6, KRYTOX 157FS(L)
      30136-13-1, Dowanol PnP
                                   68259-10-9
      ammonium salt
      RL: MOA (Modifier or additive use); USES (Uses)
         (surfactant; dilutable concs. for aqueous delivery of fluorinated
         silanes)
RE. CNT
                THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Du Pont; EP 0611771 A 1995 HCAPLUS
(2) Du Pont; WO 9523804 A 1995 HCAPLUS
(3) Minnesota Mining & Mfg; EP 0611771 A 1994 HCAPLUS
(4) Minnesota Mining & Mfg; EP 0611771 A 1997 HCAPLUS
(5) Minnesota Mining & Mfg; WO 9723432 A 1997 HCAPLUS
     ANSWER 5 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
      2003:22928 HCAPLUS
AN
      138:90650
DN
      Entered STN: 10 Jan 2003
      Imide salts as emulsifiers for the polymerization of fluoroolefins
TI
     Lamanna, William M.; Savu, Patricia M.; Sierakowski, Michael J.; Tan,
      Lian S.
PA
      3M Innovative Properties Company, USA
S<sub>0</sub>
     PCT Int. Appl., 24 pp.
     CODEN: PIXXD2
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DT
     Patent
     English
LA
     ICM C08F014-18
IC
     ICS C08F002-26
     37-3 (Plastics Manufacture and Processing)
     Section cross-reference(s): 46
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     EP 1406936
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                                 RO, MK, CY, AL,
20041014 JP 2
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     JP 2004531632
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PRAI US 2001-896319
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     WO 2002-US10291
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 WO 2003002622
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                         526/220.000; 526/217.000; 526/222.000; 526/247.000;
 US 2003018149
                 NCL
                         526/250,000; 526/254.000
                 ECLA
                         C08F014/18+2/16
 JP 2004531632
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                         4J011/KA02; 4J011/KA03; 4J011/KA04; 4J100/AC22P;
                         4J100/AC23P; 4J100/AC24P; 4J100/AC26P; 4J100/AC27P;
                         4J100/AC28P; 4J100/AC30P; 4J100/AC31P; 4J100/AE09P;
                         4J100/AE18P; 4J100/AE56P; 4J100/BA04P; 4J100/BB18P;
                         4J100/CA01; 4J100/CA04; 4J100/CA05; 4J100/FA03;
                         4J100/FA04; 4J100/FA20
OS
     MARPAT 138:90650
     The emulsifiers are comprised of imide anions that have good surface
     activity while being phys. and chemical stable during their use as
     emulsifiers. The emulsifiers of the present invention are expected to
     gradually degrade in the environment and will likely more rapidly
     bioeliminate than ammonium perfluoroctanoate. In addition, the
     emulsifiers can be produced from lower cost intermediates, such as C4F9C0F
     and C4F9S02F. Thus, bis (perfluorobutanesulfonyl) imide was
     prepared from triethylammonium bis (perfluorobutanesulfonyl) imide
     intermediate (preparation given) and used under an ammonium salt in the polymerization
     of fluoroolefins.
ST
     emulsion polymn fluoroolefin emulsifying agent fluoro
     imide salt
     Emulsifying agents
        (manufacture of imide salts as emulsifiers for polymerization of
        fluoroolefins)
IT
     Fluoropolymers, preparation
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP
     (Preparation); USES (Uses)
        (manufacture of imide salts as emulsifiers for polymerization of
        fluoroolefins)
     119229-99-1P
                    129318-48-5P
                                    482630-30-8P
IT
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP
     (Preparation); USES (Uses)
        (emulsifiers; manufacture of imide salts as emulsifiers for polymerization of
        fluoroolefins)
     81189-15-3
                  482630-27-3
                                 482630-28-4
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                                                              482630-31-9
IT
                                                 482630-35-3
     482630-32-0
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                                  482630-34-2
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RL: MOA (Modifier or additive use); USES (Uses)
         (emulsifiers; manufacture of imide salts as emulsifiers for polymerization of
         fluoroolefins)
     9011-17-0P, Hexafluoropropylene-vinylidene fluoride
                  25067-11-2P, Hexafluoropropylene-
     copolymer
     tetrafluoroethylene copolymer
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP
     (Preparation); USES (Uses)
         (manufacture of imide salts as emulsifiers for polymerization of
         fluoroolefins)
     39847-39-7P, Bis (perfluorobutanesulfonyl) imide
IT
                                                             233278-21-2P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
         (manufacture of imide salts as emulsifiers for polymerization of
         fluoroolefins)
     121-44-8, Triethylamine, reactions
                                              375-72-4,
     Perfluorobutanesulfonyl fluoride
RL: RCT (Reactant); RACT (Reactant or reagent)
         (manufacture of imide salts as emulsifiers for polymerization of
         fluoroolefins)
RE. CNT
               THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Minnesota Mining & Mfg; WO 9819988 A 1998 HCAPLUS
     ANSWER 6 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     2000:238068 HCAPLUS
     132:266766
DN
ED
     Entered STN: 13 Apr 2000
     Straight-chain and branched perfluoroalkyl halides and
     derivatives, their preparation, fluoropolymers, and use as oil-
     and water-repellant treatment agents for surfaces
     Behr, Frederick E.; Dams, Rudolf J.; Dewitte, Johan E.; Hagen,
IN
     Donald F.
     3M Innovative Properties Company, USA
PA
     U.S., 22 pp., Cont.-in-part of U.S. Ser. No. 723,049, abandoned.
     CODEN: USXXAM
DT
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LA
     English
IC
     ICM C08F018-20
INCL 526245000
     45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes)
     Section cross-reference(s): 40, 46
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                                                 US 1997-794798
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     US 6048952
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                                                 JP 2001-204928
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     US 1995-476954
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                  ECLA
                          C08G018/08D2E2; C08G018/28D8C; C08G018/80H4F
                          560/147.000; 560/152.000; 560/223.000; 562/113.000; 562/605.000; 564/209.000; 568/683.000; 568/842.000 C07C017/00+19/16; C07C053/21; C07C069/653; C07C303/24;
 US 6365769
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                          C07C303/32; C07C319/02; C07C319/14; C07C319/18;
                          CO7D303/08; C08G018/08D2E2; C08G018/09G; C08G018/28D8C; C08G018/80H4F; C08G063/682; C07C017/00+19/01;
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C07C017/00+19/14; C07C017/25+21/18; C07C019/10;
                        C07C019/14; C07C019/16; C07C031/40; C07C051/285+53/21
AB
     Aqueous treating agents for a substrate comprise compds. with pendant saturated
     perfluoroalkyl groups, where some of the perfluoroalkyl
     groups are straight chain and some are branched chain; and applying the
     polymer to the substrate; where 60-90% of the perfluoroalkyl
     groups are straight chain and . apprx. 10-40% of the perfluoroalkyl
     groups are branched chain.
     perfluoroalkyl halide surface treatment agent; straight branched
     perfluoroalkyl halide mixt; fluoropolymer
     perfluoroalkyl prepn use; oil waterproofing agent fabric
     Surfactants
        (anionic-nonionic; straight-chain and branched perfluoroalkyl
        halides and derivs. for use as oil- and water-repellent treatment
        agents for fabrics and other surfaces)
IT
     Textiles
        (cotton-polyester, substrate; straight-chain and branched
        perfluoroalkyl halides and derivs. for use as oil- and
        water-repellent treatment agents for fabrics and other surfaces)
     Polyurethanes, uses
     Polyurethanes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (fluorine-containing, perfluoroalkyl group-containing;
        straight-chain and branched perfluoroalkyl halides and
        derivs. for use as oil- and water-repellent treatment agents for
        fabrics and other surfaces)
     Polyamides, uses
IΤ
     Polycarbodiimides
     Polyethers, uses
     Polyolefins
     Polyurethanes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
         (perfluoroalkyl group-containing; straight-chain and branched
        perfluoroalkyl halides and derivs. for use as oil- and
        water-repellent treatment agents for fabrics and other surfaces)
     Fluoropolymers, uses
ΙT
       Fluoropolymers, uses
       Fluoropolymers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polyurethane-, perfluoroalkyl group-containing; straight-chain and branched perfluoroalkyl halides and derivs. for use as
        oil- and water-repellent treatment agents for fabrics and other
        surfaces)
IT
     Oilproofing agents
     Polymerization
     Textiles
     Water-resistant materials
         (straight-chain and branched perfluoroalkyl halides and
        derivs. for use as oil- and water-repellent treatment agents for
        fabrics and other surfaces)
IT Carpets
     Ceramics
     Paper
        (substrate; straight-chain and branched perfluoroalkyl
        halides and derivs. for use as oil- and water-repellent treatment
        agents for fabrics and other surfaces)
     Glass, miscellaneous
IΤ
     Metals, miscellaneous
     Plastics, miscellaneous RL: MSC (Miscellaneous)
        (substrate; straight-chain and branched perfluoroalkyl
        halides and derivs. for use as oil- and water-repellent treatment
        agents for fabrics and other surfaces)
     27854-31-5P
                                                38436-14-5P
                   27905~45-9P
                                  30389-25-4P
                                                               38436-18-9P
IT
     38565-53-6P
                   52591-27-2P
                                  81190-28-5P
                                                150940-84-4P
     RL: IMF (Industrial manufacture); PREP (Preparation)
         (straight-chain and branched perfluoroalkyl halides and
        derivs. for use as oil- and water-repellent treatment agents for
        fabrics and other surfaces)
     423-60-9P, Perfluorooctanesulfonyl chloride
                                                      423-62-1P,
IT
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865-86-1P 1693-71-6P
      Perfluorodecyliodide 678-39-7P
      Triallyl borate 2043-47-2P 2043-54-1P
                                                          34143-74-3P
                                                                             34451-25-7DP.
      reaction products with propargyl alc., phosphate esters, ammonium salts 34451-28-0P 80233-96-1P 133299-39-5P 150940-83-3P 218462-37-4P
      218462-40-9P
      RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
      (Reactant or reagent)
          (straight-chain and branched perfluoroalkyl halides and
          derivs. for use as oil- and water-repellent treatment agents for
      fabrics and other surfaces)
107-19-7DP, Propargyl alcohol, reaction products with
IT
      perfluoroalkylethanethiols, phosphate esters, ammonium salts
      107-19-7DP, Propargyl alcohol, reaction products with perfluoroalkylthiols and PAPI 9003-11-6DP, mono[ω-[(
      heptadecafluorodecyl) thio alkyl] ethers 9016-87-9DP, PAPI,
      reaction products with perfluoroalkyl alcs. 27905-45-9DP,
      reaction products with PAPI 34143-74-3DP, reaction products with
      methoxymethylmelamines, propargyl alc. and PAPI 34451-25-7DP, reaction products with propargyl alc., phosphate esters, ammonium salts 54949-95-0P 58228-15-2P 62097-34-1DP, Ethylene glycol-PAPI copolymer,
      reaction products with perfluoroalkylethanols 62880-96-0P 62880-97-1P 63225-57-0P 99332-32-8P 118570-75-5P 149759-83-1P
      150940-87-7P
                       150944-46-0P 150944-47-1P 150953-92-7P 150956-37-9P
      189398-01-4DP, phosphate esters, ammonium salts 218462-55-6P
      218462-56-7P 218462-57-8P 218462-58-9DP, reaction products with propane sultone 218462-59-0P 218462-60-3P 218462-61-4P 218462-62-5P 218462-64-7P 218462-65-8DP, reaction products with
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      propane sultone
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      218462-69-2P 218605-22-2P
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          (straight-chain and branched perfluoroalkyl halides and
          derivs. for use as oil- and water-repellent treatment agents for
          fabrics and other surfaces)
      62-56-6, Thiourea, reactions
                                           74-85-1, Ethene, reactions 107-15-3,
      Ethylene diamine, reactions 107-18-6, 2-Propen-1-ol, 109-55-7 307-51-7, Perfluorodecanesulfonyl fluoride
                                         107-18-6, 2-Propen-1-ol, reactions
      814-68-6, Acryloyl chloride 3089-11-0D, Hexamethoxymethyl melamine,
      reaction products with perfluoroalkylethylthiols 6915-15-7, Malic acid 7553-56-2, Iodine, reactions 10043-35-3, Boric acid,
                    7553-56-2, Iodine, reactions
                                                    40630-30-6 55591-23-6,
      reactions
                    15214-89-8
                                    32779-61-6
      Perfluorohexanesulfonyl chloride
      RL: RCT (Reactant); RACT (Reactant or reagent)
          (straight-chain and branched perfluoroalkyl halides and
          derivs. for use as oil- and water-repellent treatment agents for
         fabrics and other surfaces)
                THERE ARE 53 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE. CNT 53
RE
(1) Ahlbrecht; US 2764602 1956 HCAPLUS
(2) Ahlbrecht; US 2803656 1957 HCAPLUS
(3) Anon; GB 904263 1962 HCAPLUS
    Anon; EP 0142041 A1 1985 HCAPLUS
(5) Anon; JP 61209286 1986 HCAPLUS
(6) Anon; JP 6238419 1987
(7) Anon; EP 0275771 1988 HCAPLUS
(8) Anon; GB 2199828 1988 HCAPLUS
(9) Anon; JP 6445411 1989
(10) Anon; JP 532712 1993
(11) Anon; 3M Company trade bulletin 98-0211-2213-4(38.3) BPH 1988 (12) Anon; Journal of Fluorine Chemistry 1989, V43(2), P291 (13) Anon; The Journal of Organic Chemistry 1958, V23, P1166
(14) Anon; The Journal of Organic Chemistry 1988, V53(24), P5714
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(19) Billenstein; US 4167639 1979 HCAPLUS
(20) Brice; US 2732398 1956 HCAPLUS
(21) Brown; US 2759019 1956 HCAPLUS
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(22) Brown; US 2950317 1960 HCAPLUS
(23) Bultman; US 4484990 1984 HCAPLUS
     Caporiccio; US 5350878 1994 HCAPLUS
(25) Chang; US 3540126 1970 HCAPLUS
(26) Chang; US 4540497 1985 HCAPLUS
(27) Chang; US 4668406 1987 HCAPLUS
(28) Day; US 3283012 1966 HCAPLUS
(29) Dear; US 4158672 1979 HCAPLUS
(30) Diesslin; US 2567011 1951 HCAPLUS
(31) Falk; US 5091550 1992 HCAPLUS
(32) Falk; US 5132445 1992 HCAPLUS
(33) Feiring; US 5260492 1993 HCAPLUS
(34) Feiring; US 5326917 1994 HCAPLUS
(35) Francen; US 3562156 1971
(36) Fuchikami; US 5151535 1992 HCAPLUS
(37) Guenthner; US 3398182 1968 HCAPLUS
(38) Hager; US 3532659 1970 HCAPLUS
(39) Heine; US 3094547 1963
(40) Husted; US 2666797 1954 HCAPLUS
(41) Husted; US 2691043 1954 HCAPLUS
(42) Katsushima; US 3499940 1970
(43) Katsushima; US 3525758 1970
(44) Katsushima; US 3919361 1975 HCAPLUS
(45) Krahler; US 4489006 1984 HCAPLUS
(46) Landucci; US 4024178 1977 HCAPLUS
(47) Lore; US 4127711 1978 HCAPLUS
(48) Park, J; Free-Radical Catalyzed Addition of Unsaturated Alcohols to
    Perhaloalkanes 1961, P2089 HCAPLUS
(49) Pavlik; US 3420877 1969 HCAPLUS
(50) Schwenk; US 4219681 1980 HCAPLUS
(51) Simons; US 2519983 1950 HCAPLUS
(52) Stern; US 4606737 1986 HCAPLUS
(53) Tiers; US 2972638 1961 HCAPLUS
L13
     ANSWER 7 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     1999:12326 HCAPLUS
DN
     130:83186
ED
     Entered STN: 08 Jan 1999
ΤI
     Perfluoroalkyl halides and derivatives for surface treatment
     Behr, Frederick E.; Dams, Rudolf J.; Dewitte, Johan E.; Hagen,
IN
     Donald F.
PA
     Minnesota Mining & Manufacturing Company, USA
     U.S., 22 pp., Cont.-in-part of U.S. Ser. No. 489,094, abandoned.
     CODEN: USXXAM
DT
     Patent
LA
     English
     ICM C08F018-20
INCL 526245000
     45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes)
     Section cross-reference(s): 38, 40, 46
FAN. CNT 3
     PATENT NO.
                          KIND
                                  DATE
                                               APPLICATION NO.
                                                                        DATE
                                               US 1997-794828
                                                                        19970204
     US 5852148
                                  19981222
                            A
     JP 2002138078
                                  20020514
                                               JP 2001-204928
                                                                        19920710
                            A2
PRAI US 1991-728184
                                  19910710
                            B1
     US 1994-314939
                            В3
                                  19940929
     US 1995-489094
                                  19950609
                            B2
     JP 1992-183345
                            A3
                                  19920710
CLASS
                         PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                  CLASS
                  ICM
                         C08F018-20
 US 5852148
                  INCL
                         526245000
 US 5852148
                  NCL
                         526/245.000; 428/500.000; 526/243.000
                         C07C017/00+19/14; C07C051/285+53/21; C08G018/08D2E2;
                  ECLA
                         C08G018/09G; C08G018/28D8C; C08G018/80H4F;
                         C07C017/00+19/01; C07C017/00+19/16; C07C017/25+21/18
     A method for treating a substrate, comprises: providing a substrate;
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providing a polymer comprising a plurality of pendant saturated

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perfluoroalkyl groups, wherein some of the perfluoroalkyl
     groups are straight chain and some are branched chain; and applying the
     polymer to the substrate; wherein 65-85% of the perfluoroalkyl
     groups are straight chain and about 15 to about 35% of the
     perfluoroalkyl groups are branched chain. These mixts. contain
     some compds. with a straight perfluoroalkyl group and some with a branched perfluoroalkyl group. Methods of preparation and use are
     also described.
     perfluoroalkyl halide surface treatment agent
     Surfactants
         (anionic-nonionic; perfluoroalkyl halides and derivs. for
        surface treatment)
IT
     Polyurethanes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (fluorine-containing, perfluoroalkyl group-containing;
        perfluoroalkyl halides and derivs. for surface treatment)
     Polyamides, uses
IT
     Polycarbodiimides
     Polyethers, uses
     Polyolefins
     Polyurethanes, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (perfluoroalkyl group-containing; perfluoroalkyl
        halides and derivs. for surface treatment)
IT
     Oilproofing agents
     Polymerization
     Textiles
     Water-resistant materials
        (perfluoroalkyl halides and derivs. for surface treatment)
     Fluoropolymers, uses
       Fluoropolymers, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polyurethane-, perfluoroalkyl group-containing; perfluoroalkyl halides and derivs. for surface treatment)
IT
     Carpets
     Ceramics
     Paper
        (substrate; perfluoroalkyl halides and derivs. for surface
        treatment)
ΙT
     Glass, miscellaneous
     Metals, miscellaneous
     Plastics, miscellaneous RL: MSC (Miscellaneous)
        (substrate; perfluoroalkyl halides and derivs. for surface
        treatment)
IT
     27854-31-5P
                    27905-45-9P
                                    30389-25-4P
                                                   38436-14-5P
                                                                   38436-18-9P
     38565-53-6P
                    52591-27-2P
                                    81190-28-5P
                                                   150940-84-4P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (perfluoroalkyl halides and derivs. for surface treatment)
     423-60-9P, Perfluorooctanesulfonyl chloride
                                                         423-62-1P.
                                            865-86-1P
                                                          1693-71-6P
     Perfluorodecyliodide 678-39-7P
     Triallyl borate 2043-47-2P
                                      2043-54-1P
                                                     34143-74-3P
                                                                     34451-25-7P
     34451-28-0P
                   80233-96-1P
                                   133299-39-5P
                                                    150940-83-3P
                                                                     218462-37-4P
     218462-40-9P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (perfluoroalkyl halides and derivs. for surface treatment)
     107-19-7DP, Propargyl alcohol, reaction products with
     perfluoroalkylethanethiols, phosphate esters, ammonium salts 107-19-7DP, Propargyl alcohol, reaction products with perfluoroalkylthiols and PAPI 9003-11-6DP, mono[@-[(
     heptadecafluorodecyl)thio]alkyl] ethers 9016-87-9DP, PAPI,
     reaction products with perfluoroalkyl alcs. 27905-45-9DP, reaction products with PAPI 34143-74-3DP, reaction products with
     methoxymethylmelamines, propargyl alc. and PAPI 34451-25-7DP, reaction
     products with propargyl alc., phosphate esters, ammonium salts
     54949-95-0P
                   58228-15-2P
                                    62097-34-1DP, Ethylene glycol-PAPI copolymer,
     reaction products with perfluoroalkylethanols 62880-96-0P
                    63225-57-0P 99332-32-8P 118570-75-5P
                                                                    149759-83-1P
     62880-97-1P
                     150944-46-0P
                                     150944-47-1P 150953-92-7P
     150940-87-7P
                                                                      150956-37-9P
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189398-01-4DP, phosphate esters, ammonium salts 218462-55-6P
      218462-56-7P 218462-57-8P 218462-58-9DP, reaction products with propane sultone 218462-59-0P 218462-60-3P 218462-61-4P
      218462-62-5P 218462-64-7P 218462-65-8DP, reaction products with
      propane sultone 218462-66-9P 218462-67-0P 218462-68-1P
      218462-69-2P 218605-22-2P
      RL: IMF (Industrial manufacture); TEM (Technical or engineered material
      use); PREP (Preparation); USES (Uses)
           (perfluoroalkyl halides and derivs. for surface treatment)
      62-56-6, Thiourea, reactions 74-85-1, Ethene, reactions 107-15-3, Ethylene diamine, reactions 107-18-6, 2-Propen-1-ol, reactions 109-55-7 307-51-7, Perfluorodecanesulfonyl fluoride
      814-68-6, Acryloyl chloride 3089-11-0D, Hexamethoxymethyl melamine,
      reaction products with perfluoroalkylethylthiols 6915-15-7,
      Malic acid 7553-56-2, Iodine, reactions 10043-35-3, Boric acid, reactions 15214-89-8 32779-61-6 40630-30-6 55591-23-6,
      Perfluorohexanesulfonyl chloride
      RL: RCT (Reactant); RACT (Reactant or reagent)
(perfluoroalkyl halides and derivs. for surface treatment)
                  THERE ARE 53 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Ahlbrecht; US 2764602 1956 HCAPLUS
(2) Ahlbrecht; US 2803656 1957 HCAPLUS
(3) Anon; GB 904263 1962 HCAPLUS
(4) Anon; EP 0142041 A1 1985 HCAPLUS
(5) Anon; JP 61209286 1986 HCAPLUS
(6) Anon; JP 6238419 1987
(7) Anon; EP 0275771 1988 HCAPLUS
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(10) Anon; JP 532712 1993
(15) Banitt; US 3532674 1970 HCAPLUS
(16) Banks, R; Ellis Horwood Ltd 1979, P213
(17) Berger; US 4359096 1982
(18) Bernett, M; J Phys Chem 1967, V71, P2075 HCAPLUS
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(22) Brown; US 2950317 1960 HCAPLUS
(23) Bultman; US 4484990 1984 HCAPLUS
(24) Caporiccio; US 5350878 1994 HCAPLUS
(25) Chang; US 3540126 1970 HCAPLUS
(26) Chang; US 4540497 1985 HCAPLUS
(27) Chang; US 4668406 1987 HCAPLUS
(28) Day; US 3283012 1966 HCAPLUS
(29) Dear; US 4158672 1979 HCAPLUS
(30) Diesslin; US 2567011 1951 HCAPLUS
(31) Falk; US 5091550 1992 HCAPLUS
(32) Falk; US 5132445 1992 HCAPLUS
(33) Feiring; US 5260492 1993 HCAPLUS
(34) Feiring; US 5326917 1994 HCAPLUS
(35) Francen; US 3562156 1971
(36) Fuchikami; US 5151535 1992 HCAPLUS
(37) Guenthner; US 3398182 1968 HCAPLUS
(38) Hager; US 3532659 1970 HCAPLUS
(39) Heine; US 3094547 1963
(40) Husted; US 2666797 1954 HCAPLUS
(41) Husted; US 2691043 1954 HCAPLUS
(42) Katsushima; US 3499940 1970
(43) Katsushima; US 3525758 1970
(44) Katsushima; US 3919361 1975 HCAPLUS
(45) Krahler; US 4489006 1984 HCAPLUS
(46) Landucci; US 4024178 1977 HCAPLUS
(47) Lore; US 4127711 1978 HCAPLUS
(48) Park, J; Free-Radical Catalyzed Addition of Unsaturated Alcohols to
     Perhaloalkanes 1961, P2089 HCAPLUS
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(49) Pavlik; US 3420877 1969 HCAPLUS
(50) Schwenk; US 4219681 1980 HCAPLUS
(51) Simons; US 2519983 1950 HCAPLUS
(52) Stern; US 4606737 1986 HCAPLUS
(53) Tiers; US 2972638 1961 HCAPLUS
     ANSWER 8 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
AN
     1996:476619 HCAPLUS
DN
     125:117326
ED
     Entered STN: 13 Aug 1996
     Manufacture of carbodiimides as extenders for durable water-repellent
ΤI
     compositions containing fluoroorganic chemicals
     Audenaert, Frans A.; Lens, Hugo R.
IN
     Minnesota Mining and Manufacturing Co., USA Eur. Pat. Appl., 22 pp.
PA
S<sub>0</sub>
     CODEN: EPXXDW
DT
     Patent
     English
LA
     ICM C07C267-00
IC
     ICS C07C271-20; D06M013-432; D06M015-277
     40-9 (Textiles and Fibers)
     Section cross-reference(s): 23, 43, 56, 57, 58
FAN. CNT 1
     PATENT NO.
                                                                             DATE
                            KIND
                                    DATE
                                                  APPLICATION NO.
                                                  EP 1994-118500
                                                                             19941124
PΙ
     EP 713863
                             A1
                                    19960529
                                    20000426
     EP 713863
                             B1
                                     ES, FR, GB, GR, IE, IT, LI, LU, MC, NL, PT, SE
          R: AT, BE, CH, DE, DK,
                                     19981006
                                                  US 1995-546886
                                                                             19951023
     US 5817249
                             Α
                                                  JP 1995-298450
     JP 08325220
                             A2
                                                                             19951116
                                    19961210
     CN 1131147
                                    19960918
                                                  CN 1995-119341
                                                                             19951117
                             Α
PRAI EP 1994-118500
                                     19941124
CLASS
                           PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                   CLASS
 EP 713863
                   ICM
                           C07C267-00
                           C07C271-20; D06M013-432; D06M015-277
                   ICS
                           C04B041/46; C07C267/00; C07C271/20; C09K003/18;
 EP 713863
                   ECLA
                           C14C009/00; D06M013/432; D06M015/277
                   NCL
                           252/008.610; 252/008.810; 252/008.910; 427/212.000; 427/216.000; 427/221.000; 427/222.000; 524/805.000;
 US 5817249
                           524/839.000; 524/840.000; 525/123.000; 525/129.000;
                           525/130.000; 525/131.000
                           C04B041/46; C07C267/00; C07C271/20; C09K003/18; C14C009/00; D06M013/432; D06M015/277
                   ECLA
     Carbodiimide extenders manufactured by reacting isocyanates and monofunctional
     alcs., specifically C8-60 branched aliphatic alc. in non-reactive solvents in the presence of a suitable catalyst are claimed. The isocyanates and the
     monofunctional alcs., except for the OH group, are free from isocyanate-reactive H atoms. H2O-repellent compns., useful for textiles,
     fibers, leather, paper, plastic, metals, glass, concrete, etc., comprising
     a F-containing oil which is a copolymer of a fluoroalkyl
      (meth)acrylate RfZ02CCR1:CH2 [Rf = fluoroaliph. radical; Z
     C1-10 alkylene, CH2CH(OR2)CH2; R1 = H, Me; R2 = H, Ac] and a (
     fluoroalkylsulfonamido) alkyl (meth) acrylate RfSO2NR3Z102CR1:CH2 (Rf, R1 as above; R3 = H, C1-20 alkyl; Z1 = C1-20 alkylene) as
     H20-repellent and a hydrocarbon carbodismide compound are also claimed.
     typical title composition contained an aqueous emulsion of FC 3531 (weakly cationic
     emulsion fluoropolymer) and a carbodiimide extender obtained by
     reacting MDI with Prisorine 3515 (Me-branched isostearyl alc.) in 2:1 mol.
     ratio.
     carbodiimide manuf water repellent extender; MDI addn isostearyl alc water
ST
     repellent; fluoroalkyl acrylate copolymer water repellent
     Surfactants
         (additives in water-repellent compns.; manufacture of carbodiimides as
         extenders for durable water-repellent compns. containing fluoroorg
          . chems.)
IT
     Carbodiimides
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM
      (Technical or engineered material use); PREP (Preparation); USES (Uses)
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(extenders; manufacture of carbodiimides as extenders for durable
         water-repellent compns. containing fluoroorg. chems.)
IT
     Carpets
     Concrete
     Leather
     Paper
     Water-resistant materials
     Wood
         (manufacture of carbodiimides as extenders for durable water-repellent
         compns. containing fluoroorg. chems.)
IT
     Glass, oxide
     Metals, miscellaneous
     Plastics
     Rayon, miscellaneous
     Stone
     RL: MSC (Miscellaneous)
         (manufacture of carbodiimides as extenders for durable water-repellent
         compns. containing fluoroorg. chems.)
IT
     Alcohols, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
         (C12-16-branched, Isofol 14T, reaction products with isocyanates;
         manufacture of carbodimides as extenders for durable water-repellent
         compns. containing fluoroorg. chems.)
     Alcohols, uses RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM
ΙT
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
         (aliphatic, C8-60, reaction products with isocyanates, extenders; manufacture
         of carbodimides as extenders for durable water-repellent compns.
         containing fluoroorg. chems.)
     Textiles
         (cotton, manufacture of carbodiimides as extenders for durable
         water-repellent compns. containing fluoroorg. chems.)
ΙT
     Textiles
         (linen, manufacture of carbodiimides as extenders for durable
         water-repellent compns. containing fluoroorg. chems.)
IT
     Textiles
         (nonwoven, manufacture of carbodiimides as extenders for durable
         water-repellent compns. containing fluoroorg. chems.)
IT
     Amines, uses
     Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); TEM
      (Technical or engineered material use); PREP (Preparation); USES (Uses)
         (reaction products, with isocyanates, extenders; manufacture of
         carbodiimides as extenders for durable water-repellent compns. containing
         fluoroorg. chems.)
     73018-93-6
IT
     RL: MOA (Modifier or additive use); TEM (Technical or engineered material
     use); USES (Uses)
         (additive in water-repellent composition; manufacture of carbodiimides as
         extenders for durable water-repellent compns. containing fluoroorg
           chems.)
IT
     75-13-8DP, Isocyanic acid, esters, reaction products with branched C8-60
              100-37-8DP, reaction products with isocyanates 101-68-8DP, products with branched fatty alcs. 104-76-7DP, 2-Ethylhexanol,
     reaction products with branched fatty alcs.
                                               2425-77-6DP, Isofol 16, reaction
     reaction products with isocyanates
                                    2425-77-6DP, Guerbitol 16, reaction products
     products with isocyanates
                          5333-42-6DP, reaction products with isocyanates
     with isocyanates
     9004-74-4DP, Polyethylene glycol methyl ether, reaction products with isocyanates 9016-87-9DP, Voranate M 220, reaction products with branched
     fatty alcs. 26471-62-5DP, TDI, reaction products with branched fatty alcs. 27458-93-1DP, Prisorine 3515, reaction products with MDI 58670-89-6DP, Isofol 24, reaction products with isocyanates
     156930-12-ODP, Jeffamine M 715, reaction products with isocyanates
     179606-94-1DP, Guerbitol 18, reaction products with isocyanates 179606-95-2DP, Guerbitol 20, reaction products with isocyanates 179607-29-5DP, Isofol 18T, reaction products with isocyanates
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
         (manufacture of carbodiimides as extenders for durable water-repellent
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compns. containing fluoroorg. chems.)
                             141910-75-0, FC 217 1796
179607-21-7, Reaknitt DFL
     138789-49-8, FC 461
179606-90-7, FX 889
                                                      179606-89-4, FC 3531
IT
      RL: MOA (Modifier or additive use); TEM (Technical or engineered material
      use); USES (Uses)
         (manufacture of carbodiimides as extenders for durable water-repellent
         compns. containing fluoroorg. chems.)
     ANSWER 9 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
L13
      1995:829912 HCAPLUS
AN
DN
      123:343448
      Entered STN: 04 Oct 1995
ED
     Fluorosurfactants for paint and coatings
ΤI
      Witte, J. De; Piessens, G.; Dams, R.
AU
     CHEMICAL GROUP, 3M BELGIUM, Belg.
CS
      FATIPEC Congress (1994), 22nd (Vol. 2), 169-76
S0
      CODEN: FAPVAP; ISSN: 0430-2222
     Federation d'Associations de Techniciens des Industries des Peintures,
Vernis, Emaux et Encres d'Imprimerie de l'Europe Continentale
PB
DT
      Journal
      English
LA
CC
      42-5 (Coatings, Inks, and Related Products)
     Fluorosurfactants are amphiphilic materials consisting of a
      hydrophobic and oleophobic perfluorinated organic segment and a
      hydrophilic headgroup. This perfluorinated segment makes them extremely effective and efficient in lowering the surface tension of organic
      or aqueous systems. Fluorosurfactants can help to avoid typical
      paint and coating deficiencies such as crater formation, orange peeling,
      leveling and spreading, dewetting and so on.
      fluorosurfactant paint coating
      Coating materials
        Surfactants
          (fluorosurfactants for paint and coatings)
      Organic compounds, uses
      RL: MOA (Modifier or additive use); USES (Uses)
         (fluoro, fluorosurfactants for paint and coatings)
 IT
      Coating materials
         (paints, fluorosurfactants for paint and coatings)
      ANSWER 10 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
L13
      1995:708307 HCAPLUS
DN
      123:116285
      Entered STN: 29 Jul 1995
ED
      Fluorinated aliphatic radical-containing anionic sulfonamides as
ΤI
      surfactants
IN
      Dams, Rudolf J.
      Minnesota Mining and Manufacturing Co., USA
PA
      PCT Int. Appl., 29 pp.
      CODEN: PIXXD2
DT
      Patent
      English
LA
 IC
      ICM C07C311-09
      ICS C07D295-22; C07F007-08; C07C323-52; C11D001-12; C11D001-10
      46-3 (Surface Active Agents and Detergents)
      Section cross-reference(s): 23, 28
FAN. CNT 1
                                                 APPLICATION NO.
                                                                          DATE
      PATENT NO.
                            KIND
                                   DATE
                                                                          19930921
                                   19940623
                                                 WO 1993-US8863
      WO 9413634
                             A1
                           DE, DK, ES, FR, GB,
                                                GR, IE, IT, LU, MC, NL, PT, SE
          RW: AT,
                   BE, CH,
                                                 US 1992-986648
                                                                          19921208
                                    20010313
      US 6201122
                             B1
      EP 669910
                                    19950906
                                                EP 1993-922704
                                                                          19930921
                             A1
      EP 669910
                             B1
                                    19980311
          R: DE
                                    19921208
 PRAI US 1992-986648
      WO 1993-US8863
                                    19930921
 CLASS
                           PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                   CLASS
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ICM

WO 9413634

C07C311-09

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ICS
                         C07D295-22; C07F007-08; C07C323-52; C11D001-12;
                         C11D001-10
                  NCL
 US 6201122
                         544/383.000; 560/150.000; 562/106.000; 562/556.000
                  ECLA
                         CO7C311/09; CO7C323/52; CO7D295/22C2; CO7F007/08D4H4;
                         C11D001/00C
     Low-foaming surfactants such C8F17S02NHCH2CH2NHCH2CH2CONHCMe2CH2S03K (I).
AB
     C8F17S02NHCH2CH2NHCH2CH2CO2Na, C8F17S02ZCH2CH2CONHCMe2CH2S03K (Z =
     piperazine-1,4-diyl), an ethylene oxide-propylene oxide copolymer containing 1
     C8F17S02NHCHMeCH2 end group and 1 CH2CHMeNHCH2CH2CONHCMe2CH2S03K end group, C8F17S02NEtCH2CH2SCH(C02NH4) CH2CO2NH4, and C4F9S02NH(CH2)3SiMe2OSiMe2(CH2)3NHCH2CH2CONHCMe2CH2S03K are prepared by the
     Michael addition of a fluoroalkyl group-containing compound to an
     ethylenic compound The Michael addition of C8F17SO2NHCH2CH2NH2 to K
     2-acrylamido-2-methylpropanesulfonic acid gave I which formed 0.01% and
     0.05% aqueous solns. showing surface tension 18.3 and 17.1 dynes/cm, resp.
     fluoroalkylsulfonamide deriv anionic surfactant; sulfonate
     fluoroalkylsulfonamido deriv surfactant; carboxylate
     fluoroalkylsulfonamido deriv surfactant;
     acrylamidomethylpropanesulfonic reaction sulfonamide surfactant; foam redn
     anionic surfactant fluoroalkylsulfonamide
     Sulfonamides
IT
     RL: IMF (Industrial manufacture); PRP (Properties); PREP (Preparation)
         (anionic and fluoroalkyl group-containing; preparation of
        surface-active low-foaming)
     Carboxylic acids, preparation
IT
     Sulfonic acids, preparation
     RL: IMF (Industrial manufacture); PRP (Properties); PREP (Preparation)
         (fluoroalkylsulfonamido group-containing; preparation of
        surface-active low-foaming)
ΙT
     Surfactants
         (anionic, fluoroalkylsulfonamido group-containing; preparation of
         low-foaming)
     13406-94-5, N-(2-Aminoethyl) perfluoroctanesul fonamide
ΙT
                                  165527-53-7 166582-88-3
     113584-32-0
                    165527-52-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
         (addition reaction with ethylenic compds. containing anionic groups)
     6915-18-0, 2-Butenedioic acid 7446-81-3 52
RL: RCT (Reactant); RACT (Reactant or reagent)
                                      7446-81-3 52825-28-2
         (addition reaction with fluoroalkylsulfonamido group-containing
        amines)
     165527-27-5P
165527-32-2P
                     165527-28-6P
IT
                                     165527-29-7P
                                                      165527-30-0P
                                                                      165527-31-1P
                     165527-33-3P
                                     165527-34-4P
                                                     165527-35-5P
                                                                      165527-36-6P
                                                      165527-40-2P
     165527-37-7P
                     165527-38-8P
                                     165527-39-9P
                                                                      165527-41-3P
     165527-42-4P
                     165527-43-5P
                                     165527-44-6P
                                                     165527-45-7P
                                                                      165527-46-8P
     165527-47-9P
                     165527-48-0P
                                     165527-49-1P
                                                                      165527-51-5P
                                                     165527-50-4P
     166338-13-2P
                     166582-84-9P
                                     166582-85-0P
                                                     166582-86-1P
                                                                      166582-87-2P
     RL: IMF (Industrial manufacture); PRP (Properties); PREP (Preparation)
         (surfactants; preparation of low-foaming)
     ANSWER 11 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
     1994:136059 HCAPLUS
AN
     120:136059
DN
     Entered STN: 19 Mar 1994
ED
     Perfluoroalkyl halides and derivatives as precursors for oil and
     water repellants and surfactants
     Behr, Frederick E.; Dams, Rudolf J.; DeWitte, Johan E.; Hagen,
     Donald F.
     Minnesota Mining and Manufacturing Co., USA
     Can. Pat. Appl., 67 pp.
     CODEN: CPXXEB
DT
     Patent
LA
     English
     ICM C08L027-12
          C09D004-00; C09D127-12; C09D175-04; C08L075-04
     37-2 (Plastics Manufacture and Processing)
     Section cross-reference(s): 23, 40, 46
·FAN. CNT 3
     PATENT NO.
                           KIND
                                               APPLICATION NO.
                                                                        DATE
                                  DATE
                                               CA 1992-2071596
                                                                        19920618
PΙ
     CA 2071596
                            AA
                                  19930111
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```
EP 526976
                                   19930210
                                                 EP 1992-305710
                                                                           19920622
                             A1
     EP 526976
                             B1
                                   19970115
         R: BE, CH, DE, FR, GB, IT, LI, NL
05345732 A2 19931227
     IP 05345732
                                                 JP 1992-183345
                                                                           19920710
     JP 3231844
                             B2
                                   20011126
     JP 2002138078
                                                                           19920710
                                   20020514
                                                 JP 2001-204928
                             A2
PRAI US 1991-728184
                                    19910710
                             A
     JP 1992-183345
                             A3
                                   19920710
PATENT NO.
                   CLASS
                          PATENT FAMILY CLASSIFICATION CODES
CA 2071596
                   ICM
                          C08L027-12
                          C09D004-00; C09D127-12; C09D175-04; C08L075-04
C07C017/00+19/14; C07C017/00+19/01; C07C017/00+19/16;
                   ICS
EP 526976
                   ECLA
                          C07C017/25+21/18; C07C019/10; C07C019/14; C07C019/16;
                          C07C031/40; C07C051/285+53/21; C07C053/21; C07C069/653;
                          C07C303/24; C07C303/32; C07C319/02; C07C319/14; C07C319/18; C07D303/08; C08G018/08D2E2; C08G018/09G;
                          C08G018/28D8C; C08G018/80H4F; C08G063/682
     MARPAT 120:136059
     The title compds. comprise a mixture of straight and branched
     perfluoroalkyl groups bonded to Cl, Br, or I through a F-free
     alkylene group. Perfluorodecyltetrahydroiodide (prepared from
     perfluorosulfonyl fluoride, 40% straight and 60%
     branched, treated first with I, then with C2H4) was derivatized to thiol
     functionality by treatment with thiourea in EtOH to give perfluorodecyltetrahydrothiol (I). I was added to a reaction
     mixture containing hexamethoxymethylmelamine to give a I-melamine condensate
     (II, 1:4 mol ratio). A 50/50 polyester/cotton fabric blend was treated with an emulsion of II at 0.3%, dried and cured at 150°, to give a fabric with oil resistance (AATCC 118-1975) 5 and 5 after 1 dry cleaning,
     vs. 3 and 2, resp., for a precursor perfluorodecyltetrahydroiodide
     having all straight chain perfluoroalkyl groups.
     perfluoroalkyl halide prepn deriv;
     perfluorodecyltetrahydroiodide reaction thiourea; melamine
     perfluorothiol condensate treatment fabric; water repellent
     perfluoroalkyl halide deriv; oil repellent perfluoroalkyl
     halide deriv; surfactant perfluoroalkyl halide deriv
     Water-resistant materials
IT
         (fluorochem. compds. containing perfluoroalkyl groups,
         for textiles)
     Surfactants
         (amphoteric, fluorochem. compds. containing linear and branched
        perfluoroalkyl groups, preparation of)
     Surfactants
         (anionic, fluorochem. compds. containing linear and branched
        perfluoroalkyl groups, preparation of)
ΙT
     Textiles
         (cotton-polyester, water repellent agents for, chlorochem. intermediate
         as, containing perfluoroalkyl groups)
IT
     Polyoxyalkylenes, preparation
     RL: PREP (Preparation)
         (fluorine-containing, containing linear and branched
        perfluoroalkyl groups, preparation of, for manufacture of nonionic
IT
     Surfactants
         (nonionic, fluorochem. compds. containing linear and branched
         perfluoroalkyl groups, preparation of)
IT
     Fluoropolymers
     RL: PREP (Preparation)
         (polyoxyalkylene-, containing linear and branched perfluoroalkyl
         groups, preparation of, for manufacture of nonionic surfactants)
     109-55-7
ΙT
                 111-40-0, Diethylene triamine 112-24-3, Triethylene tetramine
     RL: USES (Uses)
         (linear and branched, Michael addition of, with
        perfluoroalkyltetrahydroacrylate)
     678-39-7P
IT
     RL: PREP (Preparation)
         (linear and branched, preparation and conversion of to acrylate)
                                                 27905-45-9P
                                                                30389-25-4P
                   2043-47-2P
IT
     865-86-1P
                                 27854-31-5P
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34451-28-0P
                                                 38565-53-6P
                                                               52591-27-2P
     34143-74-3P
                   34451-25-7P
     80233-96-1P
                   81190-28-5P
     RL: PREP (Preparation)
        (linear and branched, preparation of)
IT
     150944-47-1P
     RL: PREP (Preparation)
        (linear and branched, preparation of, as anionic nonionic surfactant)
     9003-11-6DP, thioethers with tetrahydroperfluorodecanethiol
     34143-74-3DP, reaction products with ethylene oxide-propylene oxide
                58228-15-2P
                               150997-16-3P
     copolymer
     RL: PREP (Preparation)
        (linear and branched, preparation of, as nonionic surfactant)
                   150909-46-9P
IΤ
     150909-45-8P
                                   150940-87-7P
     RL: PREP (Preparation)
        (linear and branched, preparation of, as surfactant)
     107-19-7DP, 2-Propyn-1-ol, reaction products with
     perfluorooctyltetrahydrothiol, urethane acrylate derivative
     678-39-7DP, oligomeric urethane derivative
                                                  3089-11-0DP, Hexamethoxymethyl
     melamine, reaction product with perfluorooctyltetrahydrothiol
     9016-87-9DP, Polymethylene polyphenylene isocyanate, reaction product with
     perfluorooctyltetrahydro alc. 26471-62-5DP, TDI, reaction
     product with propargyl alc. adduct with perfluorooctyltetrahydrothiol
        27905-45-9DP, urethane acrylate derivative 34143-74-3DP, reaction
     products with propargyl alc., urethane acrylate derivative
     RL: PREP (Preparation)
        (linear and branched, preparation of, as treatment agent for fibers for
        water resistance)
     2043-53-0P 2043-54-1P
                               2043-55-2P
                                             2043-57-4P
IT
     RL: PREP (Preparation)
        (linear and branched, preparation of, derivs. from)
    9004-74-4DP, Polyethylene glycol methyl ether, Michael adduct with
                                       54949-95-0P
     perfluoroalkyltetrahydrothiol
                                                      149759-83-1P
    150940-85-5DP, Michael adduct with perfluoroalkyltetrahydrothiol 150944-46-0P 150953-92-7DP, Michael adduct with
     perfluoroalkyltetrahydrothiol
                                       150956-33-5P
     RL: PREP (Preparation)
        (linear and branched, preparation of, for surfactant)
    110-17-8DP, 2-Butenedioic acid (E)-, reaction products with
     perfluorobutylethyl acrylate and AMPS 111-40-0DP, reaction
     products with perfluorobutylethyl acrylate and AMPS
     112-24-3DP, reaction products with perfluorobutylethyl acrylate
               52591-27-2DP, reaction products with ethylenediamine and AMPS
                                                148390-66-3DP, reaction products
     63225-57-0P
                  93776-20-6P 93857-44-4P
     with perfluorobutylethyl acrylate and ethylenediamine
                                   150953-94-9P 150956-34-6P 150956-35-7
151030-75-0DP, phosphate ester, ammonium
     149790-22-7P
                   150940-86-6P
                                                                   150956-35-7P
     150956-36-8P
                    150956-37-9P
            153326-51-3DP, phosphate ester, ammonium salts
     salts
     RL: PREP (Preparation)
        (linear and branched, preparation of, for surfactants)
     423-60-9, Perfluorooctanesulfonyl chloride 423-62-1,
     Perfluorodecyliodide 32779-61-6
                                          38436-14-5 40630-30-6
    55591-23-6, Perfluorohexanesulfonyl chloride 150940-83-3 150940-84-4
                                                      133299-39-5
     RL: USES (Uses)
        (linear and branched, reaction of, with ethylene)
     307-35-7, Perfluorooctanesulfonyl fluoride
                                                     307-51-7,
     Perfluorodecanesulfonyl fluoride
                                           375-72-4,
     Perfluorobutanesulfonyl fluoride
                                           423-50-7,
     Perfluorohexanesulfonyl fluoride
     RL: USES (Uses)
        (linear and branched, reaction of, with iodine and ethylene)
    74-85-1, Ethylene, reactions
RL: RCT (Reactant); RACT (Reactant or reagent)
IT
        (linear and branched, reaction of, with perfluoroalkyl
        iodide)
    814-68-6, Acryloyl chloride RL: USES (Uses)
IT
        (linear and branched, reaction of, with perfluoroalkyltetrahydro**
            alc.)
     107-15-3, Ethylene diamine, reactions
ΙŢ
```

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RL: RCT (Reactant); RACT (Reactant or reagent)
         (linear and branched, reaction of, with ***perfluorooctyltetrahydroac
         rylate)
     ANSWER 12 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
L13
AN
     1994:108703 HCAPLUS
DN
     120:108703
ED
     Entered STN: 05 Mar 1994
     Halogenated blowing agent emulsions and their use in manufacture of
ΤI
     fine-celled plastic foams
     Dams, Rudolf J.; Flynn, Richard M.; Focquet, Koen; Owens, John
IN
PA
     Minnesota Mining and Manufacturing Co., USA
S0
     PCT Int. Appl., 41 pp.
     CODEN: PIXXD2
DT
     Patent
LA
     English
     C08L009-14; C08L075-04
IC
CC
     37-6 (Plastics Manufacture and Processing)
FAN. CNT 1
     PATENT NO.
                          KIND
                                 DATE
                                                                     DATE
                                             APPLICATION NO.
     WO 9307201
                           A2
                                 19930415
                                              WO 1992-US7225
                                                                     19920826
PI
                                 19930513
     WO 9307201
                           A3
                 BR, CA,
         W: AU,
                          JP, KR
         RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE 5210106 A 19930511 US 1991-771442 1991
                                                                     19911004
     US 5210106
                                             US 1992-918207
     US 5211873
                                 19930518
                                                                     19920721
                           Α
                                             AU 1992-25188
                                                                     19920826
     AU 9225188
                           A1
                                 19930503
                                             EP 1992-918868
     EP 606252
                           A1
                                 19940720
                                                                     19920826
         R: BE, CH; DE,
                          ES, FR, GB, IT, LI, NL, SE
                                              JP 1992-506887
     JP 06511274
                                 19941215
                                                                     19920826
                           T2
     BR 9206579
                           A
                                 19950530
                                             BR 1992-6579
                                                                     19920826
PRAI US 1991-771442
                           Α
                                 19911004
     US 1992-918207
                           A
                                 19920721
                                 19920826
     WO 1992-US7225
                           A
CLASS
                        PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
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 WO 9307201
                  IC
                         C08L009-14IC
                                          C08L075-04
                         521/110.000; 252/182.240; 252/182.270; 521/131.000
 US 5210106
                 NCL
                         252/182. 240; 252/182. 270; 516/012. 000; 516/DIG. 001;
 US 5211873
                 NCL
                         521/112.000; 521/114.000; 521/131.000
     MARPAT 120:108703
     Emulsions for manufacture of, e.g. fine-celled polyurethane foams, contain
AB
     ≥1 high-mol.-weight compound with ≥2 reactive H atoms, ≥1
     low-boiling chlorohydrocarbon blowing agent, ≥1 Cl-free
     perfluorinated compound, a fluoro surfactant, and,
     optionally, a silicone surfactant and a hydrocarbon blowing agent. Thus,
     PAPI 135 was rapidly mixed with a mixture containing polypropoxylated sorbitol,
     water, oligomeric fluoro surfactant, N, N-
     dimethylcyclohexylamine, 2, 2-dichloro-1, 1, 1-trifluoroethane, and
     perfluoropentane to give a rigid foam with 90% fine-closed cell
     content 90%, and thermal conductivity 21.6-21.7 mW/m K.
     blowing agent emulsion polyurethane foam manuf; thermal insulator
     polyurethane foam; fluoropentane blowing agent polyurethane
     foam; fluoro surfactant blowing agent emulsion;
     chlorotrifluoroethane blowing agent emulsion polyurethane foam;
     polyoxypropylene sorbitol polyurethane foam blowing agent; PAPI
     polyurethane foam blowing agent emulsion
     Hydrocarbons, uses
       Perfluoro compounds
     RL: USES (Uses)
```

(blowing agents, emulsions containing, for manufacture of fine-celled

(halogenated, emulsions containing, for manufacture of fine-celled polyurethane

(fluoro, for halogenated blowing agent emulsions in manufacture of

polyurethane foams)

fine-celled polyurethane foams)

Surfactants

Blowing agents

IT

IT

Search done by Noble Jarrell

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foams)
    Urethane polymers, preparation RL: PREP (Preparation)
IT
        (manufacture of cellular, halogenated blowing agent emulsions in)
IT
     Thermal insulators
        (polyurethane foams, manufacture of, halogenated blowing agent emulsions in)
     Siloxanes and Silicones, uses
     RL: USES (Uses)
        (surfactants, for halogenated blowing agent emulsions in manufacture of
        fine-celled polyurethane foams)
    Hydrocarbons, uses
     RL: USES (Uses)
        (chloro, blowing agents, emulsions containing, for manufacture of fine-celled
        polyurethane foams)
IT
    Hydrocarbons, uses
     RL: USES (Uses)
        (chloro fluoro, blowing agents, emulsions containing, for manufacture
        of fine-celled polyurethane foams)
    Urethane polymers, preparation
     RL: PREP (Preparation)
        (polyoxyalkylene-, manufacture of cellular, halogenated blowing agent
        emulsions in)
     152206-27-4, Galden LS 217
IT
     RL: USES (Uses)
        (blowing agents, Galden LS-217, emulsions containing, for manufacture of
        fine-celled polyurethane foams)
    287-92-3, Cyclopentane RL: USES (Uses)
ΙT
        (blowing agents, emulsions containing halogenated blowing agents and, for
        manufacture of fine-celled polyurethane foams)
    75-29-6, Isopropyl chloride 306-83-2 311-89-7,
IT
                               335-36-4, Perfluoro
     Perfluorotributylamine
                                338-83-0, Perfluorotripropylamine
     (2-butyltetrahydrofuran)
                                  374-12-9, 1, 1, 2, 2-
74-30-1 375-83-7
     355-42-0, Perfluorohexane
                               374-30-1
     Tetrafluorocyclobutane
                                     678-26-2, Perfluoropentane
     Perfluoro-N-methylmorpholine
     1580-20-7 1600-71-1 1717-00-6
                                         49852-57-5, 1, 2, 2-Trifluoro
                                      100645-99-6
     -1-trifluoromethylcyclobutane
                                                     152076-34-1
     152076-35-2
     RL: USES (Uses)
        (blowing agents, emulsions containing, for manufacture of fine-celled
        polyurethane foams)
                   58718-85-7P 152848-87-8P
     55973-07-4P
                                               152848-88-9P
                                                                 152848-89-0P
     RL: PREP (Preparation)
        (manufacture of cellular, halogenated blowing agent emulsions in)
     79-41-4DP, siloxane esters, polymers with acrylic compds. 1120-71-4DP,
     Propane sultone, quaternary salts with tertiary amine group-containing acrylic
                                           152107-61-4P
                                                          152109-05-2DP,
              1691-99-2P 152107-60-3P
     polymer
                                           152130-86-4P
                                                           152130-87-5P
     quaternary salts with propanesultone
     152130-88-6P
     RL: PREP (Preparation)
        (manufacture of, for surfactants for halogenated blowing agent emulsions in
        manufacture of fine-celled polyurethane foams)
    ANSWER 13 OF 13 HCAPLUS COPYRIGHT 2005 ACS on STN
L13
     1993:652601 HCAPLUS
     119:252601
DN
     Entered STN: 11 Dec 1993
     Fluoro surfactants for paints and coatings: improvement of
     surface structure
     Dams, R.
     Antwerp, Belg.
     Tenside, Surfactants, Detergents (1993), 30(5), 326-7 CODEN: TSDEES; ISSN: 0932-3414
     Journal
     German
     46-4 (Surface Active Agents and Detergents)
     Section cross-reference(s): 42
     The use of fluorinated surfactants for lowering of coating
     surface tensions and thus improving the surface structure of the dried
```

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coatings was discussed.
     coating surfactant fluorinated
ST
IT
     Surfactants
         (fluorine-containing, for coatings)
IT
     Surface structure
         (of coatings, fluoro surfactants for improved)
IT
     Coating materials
         (surface structure of, fluoro surfactants for improved)
=> b wpix
FILE 'WPIX' ENTERED AT 10:52:53 ON 02 MAY 2005 COPYRIGHT (C) 2005 THE THOMSON CORPORATION
                               27 APR 2005
                                                   <20050427/UP>
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MOST RECENT DERWENT UPDATE:
                                    200527
                                                    <200527/DW>
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    FOR FURTHER DETAILS: http://www.thomsonderwent.com/dwpifv <<<
>>> THE CPI AND EPI MANUAL CODES HAVE BEEN REVISED FROM UPDATE 200501.
    PLEASE CHECK:
http://thomsonderwent.com/support/dwpiref/reftools/classification/code-revision/
    FOR DETAILS. <<<
=> d all 122 tot
     ANSWER 1 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     2004-592542 [57]
                           WPIX
     C2004-215348
     Thermally responsive composition, useful as e.g. a dental composition,
     comprises thermally responsive viscosity modifier, polymerizable component
     different than the modifier and water.
     A14 A25 A96 B05 D22
     ALI, M B; BUI, H T; LIU, J J; MA, J; MITRA, S B; NGUYEN, M T; OXMAN, J D (ALIM-I) ALI M B; (BUIH-I) BUI H T; (LIUJ-I) LIU J J; (MAJJ-I) MA J; (MITR-I) MITRA S B; (NGUY-I) NGUYEN M T; (OXMA-I) OXMAN J D; (MINN)
IN
     3M INNOVATIVE PROPERTIES CO
     107
CYC
                       A1 20040805 (200457)*
A1 20040819 (200457) EN
     US 2004151691
                                                             A61K031-765
                                                       22
                                                             A61K047-34
         2004069278
         RW: AT BE BG BW CH CY CZ DE DK EA EE ES FI FR GB GH GM GR HU IE IT KE
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          W: AE AG AL AM AT AU AZ BA BB BG BR BW BY BZ CA CH CN CO CR CU CZ DE
              DK DM DZ EC EE EG ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG
             KP KR KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NI NO NZ OM PG PH PL PT RO RU SC SD SE SG SK SL SY TJ TM TN TR TT TZ UA UG US UZ VC VN YU ZA ZM ZW
     AU 2003303893 A1 20040830 (200480)
     US 2004151691 A1 Provisional US 2003-443970P 20030130, US 2003-626261
      20030724; WO 2004069278 A1 WO 2003-US40676 20031218; AU 2003303893 A1 AU
     2003-303893 20031218
FDT AU 2003303893 A1 Based on WO 2004069278
PRAI US 2003-443970P
                             20030130; US 2003-626261
                                                                 20030724
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IC
     ICM A61K031-765; A61K047-34
          A61K006-083
     ICS
     US2004151691 A UPAB: 20040907
AB
     NOVELTY - Thermally responsive composition (I) comprises a thermally
     responsive viscosity modifier (A), a polymerizable component (B) different
     than the modifier and water.
           DETAILED DESCRIPTION - INDEPENDENT CLAIMS are also included for:
           (1) treating a surface comprising applying (I) in a low viscosity
     state at a pre-treatment temperature to the surface; and allowing the
     composition to warm to a treatment temperature and increase in viscosity
     to a highly viscous state;
           (2) hardening a composition on a surface comprising applying (I) in a
     low viscosity state at a pre-treatment temperature to the surface;
     allowing the composition to warm to a treatment temperature and increase
     in viscosity to a highly viscous state, and inducing the polymerizable
     component to polymerize; and
           (3) making a thermally responsive viscosity modifier comprising
     reacting a hydroxy-terminated poly-(oxyalkylene) polymer with an
     isocyanate-functional (meth) acrylate or a vinyl azlactone.
           ACTIVITY - Antiinflammatory.
           No biological data given.
MECHANISM OF ACTION - None given.
           USE - (I) is useful as a dental composition (suitable for use in the
     oral environment) or a medical composition (suitable for use in/on a
     body). (I) is also useful in the treatment of a surfaces and for hardening a composition on surfaces (preferably oral surface of a body such as bone, tooth, tongue, gingiva and/or throat). (All claimed.) (I) are also useful
     as e.g. reduction gels and oral coatings for hard and soft tissues; and
     also for the treatment of e.g. periodontal disease, gingivitis,
     sensitivity, halitosis and xerostomia.
           ADVANTAGE - (I) provides hardenable compositions (e.g. hardenable
     gels) that offer advantages like dimensional stability, thermal stability,
     improved stability to liquids, improved adhesion and the potential for
     sustained release of incorporated additives. The substantial moisture
     content of (I) provides the ability to easily deliver or apply a
     gel-on-contact aqueous material that provides substantial hydration of
     tissues that are subjected to dehydration.
     Dwg. 0/9
FS
     CPI
FA
     AB; DCN
     CPI: A10-E07B; A10-E24; A12-V01; A12-V02B; B04-C03; B04-L01; B05-C07;
MC
           B05-C08; B10-A13D; B14-N06; D08-A
L22
     ANSWER 2 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
                          WPIX
     2004-439058 [41]
                          DNC C2004-164427
DNN
     N2004-347350
     Aqueous etch solution for etching silicon oxide-containing substrates,
     comprises acid and surfactant containing perfluoroalkyl, hydroxyalkyl,
     alkylamine oxide, alkylcarboxylate or aminoalkyl and cation.
     E19 L03 U11
     FLYNN, R M; PARENT, M J; SAVU, P M (MINN) 3M INNOVATIVE PROPERTIES CO
IN
CYC
     106
                       A1 20040513 (200441)*
A1 20040527 (200441) EN
     US 2004089840
PΙ
                                                             C09K013-00
     WO 2004044091
                                                             C09K013-04
         RW: AT BE BG CH CY CZ DE DK EA EE ES FI FR GB GH GM GR HU IE IT KE LS
             LU MC MW MZ NL OA PT RO SD SE SI SK SL SZ TR TZ UG ZM ZW
          W: AE AG AL AM AT AU AZ BA BB BG BR BY BZ CA CH CN CO CR CU CZ DE DK
             DM DZ EC EE EG ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG KP KR KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NI NO NZ OM PG PH PL PT RO RU SC SD SE SG SK SL SY TJ TM TN TR TT TZ UA UG UZ VC
             VN YU ZA ZM ZW
     AU 2003272331 A1 20040603 (200470)
                                                             C09K013-04
     US 2004089840 A1 US 2002-290765 20021108; WO 2004044091 A1 WO 2003-US28606
     20030911; AU 2003272331 A1 AU 2003-272331 20030911
     AU 2003272331 A1 Based on WO 2004044091
FDT
PRAI US 2002-290765
                             20021108
     ICM C09K013-00; C09K013-04
IC
     ICS C09K013-08; H01L021-311
     US2004089840 A UPAB: 20040629
AB
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NOVELTY - An aqueous etch solution comprises acid and surfactant
     containing perfluoroalkyl, hydroxyalkyl, alkylamine oxide,
     alkylcarboxylate or aminoalkyl, and cation.
          DETAILED DESCRIPTION - An aqueous etch solution comprises acid and
     surfactant of formula Rf-SO2(N-)-R1M+.
          Rf = 1-12C perfluoroalkyl group;
          R1 = H, alkyl, hydroxyalkyl, alkylamine oxide, alkylcarboxylate or
     aminoalkyl;
     M+ = cation
          An INDEPENDENT CLAIM is also included for a method of etching
     comprising contacting a substrate with the etch solution.
          USE - For use in etching silicon oxide-containing substrates, e.g.
     integrated circuit.
          ADVANTAGE - The inventive etch solution includes fluorinated
     surfactant that is stable in the aqueous acid etch solution and reduces
     the surface tension so that nanoscale features may be effectively provided
     to a silicon substrate. It has same etch rate as conventional etch
     solutions and possesses low surface tension. It is non-forming, low in
     particulates that may contaminate a substrate and leaves low or no surface
     residues on rinsing. It has improved stability of performance when
     filtered or after extended storage and finally affords excellent substrate
     surface smoothness.
     Dwg. 0/0
     CPI EPI
     AB; DCN
CPI: E07-D04A; E10-A01; E10-A03B; E10-A08C; E31-B03C; L04-C07C1
FA
     EPI: U11-A10
     ANSWER 3 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
L22
                         WPIX
     2004-430593 [40]
     N2004-340480
                         DNC C2004-161135
     Aqueous etch solution for etching of silicon-containing substrate for
     microelectronic devices, e.g. integrated circuit, comprises acid and
     surfactant.
     E19 L03 U11
     FLYNN, R M; LAMANNA, W M; MOORE, G G I; PARENT, M J; QIU, Z; SAVU, P M;
     ZHANG,
     (MINN) 3M INNOVATIVE PROPERTIES CO
PA
CYC
     106
     US 2004094510
                     A1 20040520 (200440)*
A1 20040527 (200440) EN
                                                         C23F001-00
                                                  11
     WO 2004044092
                                                         C09K013-04
        RW: AT BE BG CH CY CZ DE DK EA EE ES FI FR GB GH GM GR HU IE IT KE LS
            LU MC MW MZ NL OA PT RO SD SE SI SK SL SZ TR TZ UG ZM ZW
            AE AG AL AM AT AU AZ BA BB BG BR BY BZ CA CH CN CO CR CU CZ DE DK DM DZ EC EE EG ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG KP
            KR KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NI ÑO NZ OM PG
            PH PL PT RO RU SC SD SE SG SK SL SY TJ TM TN TR TT TZ UA UG UZ VC
             VN YU ZA ZM ZW
     AU 2003278831 A1 20040603 (200470)
                                                        C09K013-04
    US 2004094510 A1 US 2002-290763 20021108; WO 2004044092 A1 WO 2003-US29262
     20030917; AU 2003278831 A1 AU 2003-278831 20030917
     AU 2003278831 Al Based on WO 2004044092
PRAI US 2002-290763
                           20021108
     ICM C09K013-04; C23F001-00
         C09K013-08; H01L021-311
     US2004094510 A UPAB: 20040624
AB
     NOVELTY - An aqueous etch solution comprises an acid and a surfactant.
          DETAILED DESCRIPTION - An aqueous etch solution comprises an acid and
     a surfactant of formula RfQR1SO3-M+.
Rf = 1-12C perfluoroalkyl group;
          R1 = alkylene of formula CnH2n(CHOH)oCmH2m, and is optionally
     substituted by a catenary oxygen or nitrogen group;
     n, m = 1-6;
     o' = 0-1;
     M+ = cation;
     Q = 0 or SO2NR2;
          R2 = H, alkyl, aryl, hydroxyalkyl, aminoalkyl, or sulfonatoalkyl
     group, optionally containing catenary oxygen or nitrogen heteroatoms.
          An INDEPENDENT CLAIM is also included for etching a substrate by
     contacting the substrate with the above etch solution.
```

```
USE - For etching of silicon-containing substrates (claimed) for
microelectronic devices, such as integrated circuit, flat panel displays;
microelectromechanical systems; or electronic equipment including personal
computers, cellular phones, electronic calendars, personal digital
assistants, or medical electronics, for televisions, stereo components, or
automobiles.
     ADVANTAGE - The fluorinated surfactant is stable in the aqueous acid
etch solution, and reduces the surface tension of the solution, thus
providing nanoscale features to the substrate. The etch solution has the
same etch rate as the conventional etch solutions and possesses low
surface tension resulting in low contact angles between the solution and
substrate. It is non-foaming, and is low in particulates that may contaminate the substrate and leaves low or no surface residues on rinse.
It provides improved stability of performance when filtered or after
extended storage and affords excellent substrate surface smoothness.
```

Dwg. 0/0 CPI EPI FA AB; DCN

CPI: E07-D04A; E10-A01; E10-A09B7; E10-A09B8; E10-A10D; E10-A22; E31-B03C; MC E32-A04; L04-C07C1

EPI: U11-A10

ANSWER 4 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN L22

WPIX 2003-310954 [30] AN

1996-354514 [35]; 1997-341674 [31]; 1999-600394 [51]; 2001-570046 [64]; 2002-238370 [29]; 2002-506442 [54]; 2002-665798 [71]; 2003-090300 [08]; CR 2003-644573 [61]; 2003-764656 [72]

C2003-081412

Cleaning composition for removing contaminants, e.g., hydrocarbons, from surface of substrates, e.g. metal, comprises mono-, di-, or trialkoxy-substituted perfluoroalkane compound(s) and surface active agent(s).

DC D25 E19

FLYNN, R M; MOORE, G G I; OWENS, J G IN (MINN) 3M INNOVATIVE PROPERTIES CO PA

CYC

PΙ US 2002169098 A1 20021114 (200330)* 18 C11D017-00 B2 20030121 (200330) US 6509309 C11D007-30

US 2002169098 A1 CIP of US 1995-375812 19950120, CIP of US 1995-573416 ADT 19951215, Div ex US 1999-268236 19990315, Div ex US 2001-867169 20010529, US 2002-96713 20020312; US 6509309 B2 CIP of US 1995-375812 19950120, CIP of US 1995-573416 19951215, Div ex US 1999-268236 19990315, Div ex US 2001-867169 20010529, US 2002-96713 20020312

US 2002169098 A1 CIP of US 5925611, Div ex US 6291417, Div ex US 6380149; US 6509309 B2 CIP of US 5925611, Div ex US 6291417, Div ex US 6380149

19990315; US 1995-375812 19950120; PRAI US 1999-268236 19951215; US 2001-867169 20010529; US 1995-573416 US 2002-96713 20020312

ICM C11D007-30; C11D017-00 IC

ICS C11D003-24

US2002169098 A UPAB: 20031107 AB

NOVELTY - Cleaning composition comprises at least one mono-, di-, or $trial koxy-substituted\ perfluoroal kane,\ perfluorocycloal kane,$ perfluorocycloalkyl-containing perfluoroalkane, or perfluorocycloalkylenecontaining perfluoroalkane compound; and at least one surface active agent.

USE - The cleaning composition is used for removing contaminants, e.g., hydrocarbons, fluorocarbons, or even water, from the surface of a

substrate, e.g. metal, glass, ceramic, plastic, or fabric.

ADVANTAGE - The alkoxy-substituted compounds used in the inventive composition exhibit unexpectedly high stabilities in the presence of acids, bases, and oxidizing agents; and in spite of their fluorine content, these compounds are good solvents for hydrocarbons and fluorocarbons. They are low in toxicity and flammability, have zero depletion potentials, and have short atmospheric lifetimes and low global warming potentials relative to chlorofluorocarbons and chlorofluorocarbon substitutes. Their good solvency and environmentally acceptable properties satisfy the need for substitutes or replacements for the commonly-used cleaning solvents which have been linked to the destruction of the earth's ozone layer.

```
Dwg. 0/0
     CPĬ
FS
     AB; DCN
FA
MC
     CPI: D11-A; D11-B; D11-D01; E07-D05; E07-E03; E10-H01; E10-H01C; E11-F05
L22
     ANSWER 5 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     2001-556747 [62]
                          WPIX
AN
CR
     1999-429973 [36]
                          DNC C2001-165524
DNN
     N2001-413666
     Electrolyte composition useful in electrochemical systems such as
ΤI
     batteries, comprises specific conductive salt and surfactant salt.
DC
     E19 L03 V01 X16
IN
     BOYD, S D; FANTA, A D; JOHNSON, B J; LAMANNA, W M; LOCH, R B; PHAM, P T;
     SHIMADA, H
PA
     (MINN) 3M INNOVATIVE PROPERTIES CO
CYC
    US 6280883 B1 20010828 (200162)* 16 H01M006-18 US 6280883 B1 CIP of US 1997-988507 19971210, US 1999-267310 19990312
PΙ
ADT
PRAI US 1999-267310
                            19990312; US 1997-988507
                                                              19971210
IC
     ICM H01M006-18
AB
     US
          6280883 B UPAB: 20011026
     NOVELTY - An electrolyte composition comprises a conductive salt, and a
     surfactant salt (IA). The conductive salt comprises alkali metal, alkaline
     earth metal, a group IIB metal, a group IIIB metal, a rare earth metal, a
     nitrogen onium cation or a proton, an anion with perfluoroalkyl sulfo
     group, a bis-fluoroalkylsulfonyl methide anion and a tris-
     (perfluoroalkanesulfonyl) methide anion.
           DETAILED DESCRIPTION - An electrolyte composition comprises
     conductive salt, and surfactant salt (IA), at the molar ratio of 99.9:0.1-75:25. The conductive salt comprises cation chosen from alkali
     metal, alkaline earth metal, group IIB metal, group IIIB metal, rare earth
     metal, nitrogen onium cation or proton, anion of formula Rf0S03-, where
     RfO is a 2-4C perfluoroalkyl group, anion (I, II), bis-fluoroalkylsulfonyl methide anion (III) and tris-(perfluoroalkanesulfonyl)methide anion (IV).
           Rf1, Rf2 = optionally branched 1-4C perfluoroalkyl group, Rf1 and
     Rf2 have a total of up to 5 carbon atoms;
           Rf3 = 2-4C perfluoroalkylene moiety optionally substituted by
     optionally branched 1-2C perfluoroalkyl group, Rf3 has total of up to 4
     carbon atoms;
           X- = 0-, N-S02Rf4 or -C(Rf6S02) (S02Rf7);
Z = CF2, 0, NRf8 or Sf4;
          Rf4, Rf5 = CmF2m+1' or (CF2)q-S02-X-;
Rf6, Rf7 = CmF2m+1', (CF2)4-S02-X-, N(Rf1) (Rf2)(CF2)n' or group
           Rf8 = CmF2m+1' or (CF2)q-S02-X-;
           Rf6', Rf7' = perfluoroalkylene moiety of formula CrF2r;
     n', r, n, q = 1-4;
RfII, RfIII = 1-4C perfluoroalkyl groups, RfI and RfII have a total
of up to 5 carbon atoms;
           R = H, CN, F, 1-6C alkyl, phenyl optionally substituted with 1-4C
           RfIII, RfIV, RfV = 1-4C perfluoroalkyl group, RfIII, RfIV, and RfV
     have total of up to 6 carbon atoms;
          M+n = cation with the valence equal to n;
        = 1-4;
                   = 1-12C perfluoroalkyl, perfluorocycloalkyl or
     perfluorocycloalkyl perfluoroalkyl having 4-7C ring carbon atoms and 1-4C
     alkyl chain and optionally containing catenary heteroatoms, Rf and Rf
     have a total of 8 carbon atoms.
           INDEPENDENT CLAIMS are also included for the following:
           (i) an electrochemical system which comprises a cathode (6), a anode
     and an electrolytic composition;
           (ii) battery or rechargeable battery comprising a
     lithium-intercalated carbon anode and a metal oxide cathode;
           (iii) a method of improving safety and performance of an
     electrochemical system which involves using an electrolyte composition
     comprising surfactant salt (II), a conductive salt having a cation, an
     anion (I), and a matrix material.
           USE - Useful in electrochemical system such as batteries, e.g.
     primary and secondary (rechargeable batteries), double-layer capacitors,
```

```
supercapacitors, fuel cells, electroplating, electrorefining systems.
          ADVANTAGE - Perfluorinated imide surfactant salts in electrolyte
     comprising short chain bis(perfluoroalkanesulfonylimide) conductive salts
     reduces exotherm energies produced at the electrode/electrolyte interface
     when a battery temperatures is up to 200 deg. C. The surfactant salts in
     electrolytes comprising short chain bis (perflouroalkanesulfonylimide)
     conductive salts maintains the high ionic conductivity, solubility, and
     chemical and thermal stability of the electrolyte. The surfactant and
     conductive salts lower the surface tension of the electrolyte solution and
     allow rapid wetting of battery component materials such as separators and
     electrodes to save time, process steps, and process equipment. The
     surfactant and conductive salts expand the range of electrolyte
     compositions available by allowing the use of high viscosity, high surface
     energy solvents. The surfactant and conductive salts increase the area of
     contact between the electrolyte and the electroactive materials to
     decrease battery internal resistance and increase performance, and are
     used in small quantity additions to existing electrolyte formulations to
     enhance performance. The surfactant and conductive salts provide for
     potential applications in a variety of battery systems including aqueous
     electrolyte batteries as well as fuel cells and capacitors. Corrosion of
     aluminum current collectors is reduced by the surfactant salt, as
     repassivation potential of the cell is increased to over 4.5 volts, thus
     reducing the corrosion current at high positive potentials. Capacity fade
     in cells during high temperature cycling and storage (e.g., 60 deg. C, or
     higher at full charge of 4.2V) is reduced by the surfactant salt.
     Surfactant salt/conductive salt gives electrolyte salt performance
     comparable to lithium hexafluorophosphate. The drawbacks of hydrolytic and
     thermal instability inherent in the hexafluorophosphate anion, in lithium
     hexafluorophosphate, are prevented.

DESCRIPTION OF DRAWING(S) - The figure shows the view of a lithium
     ion battery.
     Anode 3
     Cathode 6
     Dwg. 1/1
     CPI EPI
     AB; GI; DCN
     CPI: E05-A; E05-B01; E05-L; E05-M; E05-N; E05-P; E07-D04A; E07-D09A;
          E07-F03; E07-H; E10-A08C; E10-A09B8; E10-A10D; E10-A22; L03-E01C
     EPI: V01-B01B1; V01-B01D; X16-A02A; X16-B01F1; X16-C; X16-J01A; X16-L02
     ANSWER 6 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
     1998-286826 [25]
                         WPIX
    N1998-225441
                         DNC C1998-088857
DNN
     Fluoro-alkyl-carbonyl fluoride(s) and derivatives used as e.g. surfactants
      having one or more alpha branched fluoroalkyl carbonyl moieties.
     A60 D25 E19 G02 G06 H01 M11 P35
     FAN, W; MANZARA, A P; MOORE, G G I; STERN, R M
     (MINN) MINNESOTA MINING & MFG CO; (MINN) 3M INNOVATIVE PROPERTIES
     CO
     78
                      A1 19980514 (199825)* EN 84
     WO 9819988
                                                         C07C053-50
        RW: AT BE CH DE DK EA ES FI FR GB GH GR IE IT KE LS LU MC MW NL OA PT
            SD SE SZ UG
         W: AL AM AT AU AZ BA BB BG BR BY CA CH CN CU CZ DE DK EE ES FI GB GE
            GH HU IL IS JP KE KG KP KR KZ LC LK LR LS LT LU LV MD MG MK MN MW MX NO NZ PL PT RO RU SD SE SG SI SK TJ TM TR TT UA UG UZ VN YU
     AU 9723396
                      A 19980529 (199841)
                                                         C07C053-50
     ZA 9709420
                      A 19990630 (199931)
                                                         C07C000-00
                                                   83
                      A1 19990825 (199939)
                                                         C07C053-50
     EP 937027
                                             EN
         R: BE CH DE FR GB IT LI
     US 6013795
                         20000111 (200010)
                                                         C07D295-00
                      Α
                         19991229 (200019)
20000825 (200121)
     CN 1239940
                                                         C07C053-50
                      Α
     KR 2000052960
                                                         C07C053-50
                      A
                         20020514 (200236)
     IP 2002514190
                                                   86
                                                         C07C051-04
    WO 9819988 A1 WO 1997-US4582 19970321; AU 9723396 A AU 1997-23396
     19970321; ZA 9709420 A ZA 1997-9420 19971021; EP 937027 A1 EP 1997-916142 19970321, WO 1997-US4582 19970321; US 6013795 A US 1996-743479 19961101;
     CN 1239940 A CN 1997-180362 19970321; KR 2000052960 A WO 1997-US4582
     19970321, KR 1999-703834 19990430; JP 2002514190 W WO 1997-US4582 19970321, JP 1998-521345 19970321
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PA

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FDT AU 9723396 A Based on WO 9819988; EP 937027 A1 Based on WO 9819988; KR
      2000052960 A Based on WO 9819988; JP 2002514190 W Based on WO 9819988
PRAI US 1996-743479
                              19961101
      ICM C07C000-00; C07C051-04; C07C053-50; C07D295-00
      ICS C07C031-38; C07C031-40; C07C053-21; C07C059-135; C07C069-63;
           C07C069-653; C07C205-43; C07C219-06; C07C233-36; C07C237-00; C07C291-04; C07C309-15; C07C327-00; C07C327-20; C07D211-00;
           C07D213-81; C07D213-89; C07D265-00; C07D295-12; C07D295-14;
           C07D295-22; C11D001-00; C11D001-88; C11D001-90
ICA
     A62D001-00
      WO
           9819988 A UPAB: 19980624
AB
      A composition comprises one or more alpha -branched fluoroalkylcarbonyl
      fluoride compounds of formula (I).
           Rf-C(R'f)FC(=0)F
                                   (I)
           Also claimed, a composition which comprises a compound having one or
      more alpha -branched fluoroalkylcarbonyl moieties of formula (II).
            [Rf-C(R'f)FC(=0)-]pX
                                        (II)
           Also claimed, a composition which comprises one or more alpha
      -branched 1, 1-dihydrofluoroalkyl moieties of formula (III).
           RfC(R'f)FCH2X1
                                 (III)
           Also claimed, a composition which comprises a compound of formula
                                   1/9Mq+
           RfC(R'f)FC(=0)0-
                                                (IV)
           Also claimed, a method of treating a compound of formula (IV) by
     heating to >140 deg. C to give monohydride and fluoroolefin. In formula (I): Rf, R' f = an acyclic fluorinated group bonded through
      C that is either substituted or unsubstituted, linear or branched and may
      optionally contain one or more catenary heteroatoms; Rf or R'f may contain
      one or more -H or one or more other halogen atoms provided that at least 75% of the atoms attached to the C backbone are fluorine atoms; and at
      least 1 of Rf and R'f contain at least 5 C's where the sum of C atoms is
      at least 7.
           In formula (II): Rf, R'f = fluorinated group bonded through C that is
      either substituted or unsubstituted, cyclic or acyclic, linear or branched
      and may optionally contain one or more catenary heteroatoms. Rf or R'f may
      contain one or more -H or one or more halogen atoms provided that at least
      75% of the atoms attached to the C backbone are F atoms; p = 1 to many equalling the valency of X; X = \text{halogen}, hydroxyl, or a moiety remaining
      after the reaction of an alpha -branched fluoroalkylcarbonyl fluoride with
      a reagent containing at least 1 active H atom after the elimination of
     hydrogen fluoride; and when X = H or hydroxyl, p = 1, Rf and R'f = acyclic alkyl and at least 1 of Rf and R'f contains at least 5C atoms. The sum of
      the C atoms in Rf and R'f is at least 7. X1 = -OR, -OSO2R, -OCOR,
      -N(R1)(R2), -SR, halogen or an ethylenically unsaturated moiety; R, R1, R2
      = H, alkyl, aryl, alkaryl, aralkyl that may be substituted or
      unsubstituted, saturated or unsaturated, linear or branched, cyclic or
      acyclic and may contain one or more catenary heteroatoms where R1 and R2
      together = heterocyclic ring; Q = substituted or unsubstituted divalent
      organic group; M = cation selected from H+, metal cation, ammonium cation,
      substituted ammonium cation, polyammonium cation; and q = valency of M.
           USE - The composition can be used as surfactants and emulsifiers in
      fluoropolymer emulsions, cleaning solutions, aqueous film forming foams,
      coating additives, plating baths, wetting agents, floor polish levelling
      agents, dispersion aids, oil well stimulation chemicals, photographic
     coupling agents, soil, water and oil repellent compositions.

ADVANTAGE - The derivatives are environmentally non-persistent and
     have much lower toxicity than their linear and cyclic homologues. The salt
      compositions are low in toxicity and thermally degrade quickly at low temperatures, 80-100 deg. C, in aqueous media. They breakdown in the
      environment to volatile, non-surface active species.
      Dwg. 0/0
     CPĪ GMPI
FS
FA
     AB; DCN
     CPI: A08-S05; D11-A; E10-A25B2; G02-A05; H08-E05; M11-B
MC
     ANSWER 7 OF 7 WPIX COPYRIGHT 2005 THE THOMSON CORP on STN
L22
      1996-068625 [07]
AN
                           WPIX
     1996-020330 [02]
CR
DNN
     N1996-057739
                           DNC C1996-022275
     Omega-hydro-fluoroalkyl ether(s) useful as CFC substitutes - are
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environmentally friendly and useful, e.g. as solvents, coolants in
       refrigerators, chemical fire extinguishing agents and in synthetic blood. A60 B05 E15 E16 G04 K01 L03 M12 P35 P55
DC:
        FLYNN, R M; GUERRA, M A; MOORE, G G I; OWENS, J G; MOORE, G G
       (ASAH) ASAHI CHEM CO LTD; (BOSC) BOSCH GMBH ROBERT; (MINN) MINNESOTA MINING & MFG CO LTD; (MINN) MINNESOTA MINING & MFG CO; (MINN) 3M
PA
        INNOVATIVE PROPERTIES CO
CYC
                                A1 19951130 (199607)* EN
                                                                          63
                                                                                     C07C043-12
        WO 9532174
PΙ
            RW: AT BE CH DE DK ES FR GB GR IE IT LU MC NL PT SE
              W: CA CN JP RU
                                A1 19970312 (199715) EN
                                                                                    C07C043-12
        EP 760809
              R: DE FR GB IT
                                     19970819 (199739)
                                                                                     C09K003-00
        US 5658962
                                                                           15
                                A
                                     19980127 (199814)
        JP 10500950
                                                                                     C07C043-12
                                                                                     A62C002-00
        US 6024176
                                A
                                     20000215
                                                    (200016)
                                     19970423
                                                    (200109)
                                                                                     C07C043-12
        CN 1148377
                                Α
                                     20010320
                                                    (200118)
                                                                                     C08G018-00
        US 6204299
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        US 6214253
                                B1 20010410 (200122)
                                                                                     C09K005-00
                                     20011004 (200161)
                                                                                     C11D001-00
        US 2001027172
                                A1
        EP 1170275
                                     20020109 (200205)
                                                                                     C07C043-12
                                A2
                                                                   EN
              R: DE FR GB IT
        RU 2177934
                                     20020110 (200220)
                                                                                     C07C043-12
                                C2
        US 6361713
                                     20020326 (200226)
                                В1
                                                                                     C09K003-00
        EP 760809
                                B1
                                     20020724 (200256)
                                                                   EN
                                                                                     C07C043-12
              R: DE FR GB IT
        DE 69527527
                                     20020829 (200264)
                                                                                     C07C043-12
        US 6491983
                                B2 20021210
                                                    (200301)
                                                                                     B05D001-00
        US 2003166487
                                A1 20030904
                                                    (200359)
                                                                                     C11D017-00
                                     20040722
                                                    (200448)
        JP 2004203889
                                                                           32
                                                                                     C07C043-12
                                Α
        JP 3590067
                                B2 20041117 (200475)
B2 20050308 (200518)
                                B2 20041117
                                                                           29
                                                                                     C07C043-12
                                                                                     C07C019-08
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ICA
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      A normally liquid, omega-hydrofluoroalkyl ether cpd. of formula (I) is new:
      X = a \text{ primary H}; X = F, \text{ primary H or primary Cl; } n = 0-7; Rf, Rf', Rf'' = perfluoroalkylene (opt. substd. with perfluoroalkyl or perfluorocycloalkyl
      perfluoroalkylene; b, d 3; c 1; Z = COOH, COOM1/v, COONH4, COOR, CH2OH, COF, COC1, COR, CONRR, CH2NH2, CH2NCO, CN, CH2OSO2R, CH2OCOR, CH2OCOCR=CH2, CONH(CH2)mSi(OR)3 or CH2O(CH2)mSi(OR)3; M = ammonium or metal atom of valency v; v = 1-4; m = 1-11; R = 1-14C alkyl opt. containing a
      heteroatom, 1-14C fluoroalkyl and 6-10 membered aryl.

USE - (I) are useful in applications where CFCs, HCFCs or halons have
      been used, e.g. as solvents for precision or metal cleaning of electronic
      articles such as discs or circuit boards, in vapour phase soldering, to
      displace water from surfaces, heat transfer agents, coolants in
      refrigerator or freezer compressors or air conditioners, blowing agents or
      cell size regulators in making polyurethane foam insulation or chemical
      fire extinguishing agents in streaming applications, total flooding,
      explosion suppression and inertion and as carrier solvents (or highly fluorinated polyethers used as lubricants (or magnetic recording media.

    are also useful in various medical and oxygen transport applications

      e.g. artificial or synthetic bloods. (II) and (III) are precursors of (I).
      (II) and (III) can be converted into other derivs., e.g. the ammonium salt which can be used as surfactants, elastomers, coatings, lubricants, substances used in the preparation of liquid crystal materials and in the
      treatment of fibrous substrates to impart oil and water repellancy.
            ADVANTAGE - (I) are hydrophobic and less oleophotic than the
      perfluoroalkyl ether analogues, chemically inert, thermally stable, water insoluble and normally liquid (e.g. at 20 deg.C). They can be made in high
      yield, high purity and with a wide range of mol. wts. The covalent bond
      between the -H and the terminal C is generally degradable by atmospheric
      photo-oxidation, thus making (I) environmentally acceptable or compatible.
      (II) and (III) are more soluble in aqueous media and other organic solvents
      than are the corresponding perfluoroalkanoic acid derivs. (II) and (III)
      can also be prepared in high yields, as single molecular species.
      Dwg. 0/0
      CPI GMPI
      AB; GI; DCN
FA
      CPI: A08-B04A; B05-B01B; B10-A09B; B10-A14; B10-A15; B10-A25; B10-B03B; ·
            B10-C04D; B10-D03; B10-E04C; B10-F02; B10-G02; B10-H01;
            B12-M09; B14-F11; E10-H01C; G04-B01; G04-B08; K01-A;
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